## Article

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#### Abstract

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# Demonstrating In-Cell Target Engagement using a 

# Pirin Protein Degradation Probe (CCT367766) 

Nicola E. A. Chessum, ${ }^{\text {I\# }}$ Swee Y. Sharp, ${ }^{\text {I\# }}$ John J. Caldwell, ${ }^{1}$ A. Elisa Pasqua, ${ }^{1}$ Birgit Wilding, ${ }^{1}$ Giampiero Colombano, ${ }^{1}$ Ian Collins, ${ }^{1}$ Bugra Ozer, ${ }^{1}$ Meirion Richards, ${ }^{1}$ Martin Rowlands, ${ }^{1}$ Mark Stubbs, ${ }^{1}$ Rosemary Burke, ${ }^{1}$ P. Craig McAndrew, ${ }^{1}$ Paul A. Clarke, ${ }^{I^{*}}$ Paul Workman, ${ }^{{ }^{*}}$ Matthew D. Cheeseman ${ }^{1 *}$ and Keith Jones ${ }^{I^{*}}$<br>${ }^{1}$ Cancer Research UK Cancer Therapeutics Unit at The Institute of Cancer Research, London SW7 3RP, United Kingdom


#### Abstract

Demonstrating intracellular protein target engagement is an essential step in the development and progression of new chemical probes and potential small molecule therapeutics. However, this can be particularly challenging for poorly studied and non-catalytic proteins, as robust proximal biomarkers are rarely known. To confirm that our recently discovered chemical probe 1 (CCT251236) binds the putative transcription factor regulator pirin in living cells, we developed a heterobifunctional protein degradation probe. Focusing on linker design and physicochemical properties, we generated a highly active probe $\mathbf{1 6}$ (CCT367766) in only three iterations, validating our efficient strategy for degradation probe design against non-validated protein targets.


## INTRODUCTION

Drug discovery is reliant on recombinant proteins and biochemical screens to develop structure activity relationships (SAR) and progress compounds. ${ }^{1}$ However, the conditions in biochemical assays often display little relevance to the intracellular environment, which can result in a failure to translate high target affinity to activity within a living cell. ${ }^{2}$ Target-proximal biomarker modulation is the most important confirmation of intracellular target engagement. ${ }^{3}$ Unfortunately, this is often not possible in early stage chemical probe or drug discovery projects, especially against novel biological targets, where poorly understood biology and pharmacology make it difficult to discover and robustly validate biomarkers; a process that is particularly challenging for non-catalytic proteins.

Owing to the importance of confirming intracellular target engagement, several techniques have been developed. The over-expression of fusion proteins allows for a direct readout of target occupancy. ${ }^{4}$ However, the engineered cell lines are often difficult to generate and the protein label can impact compound binding. The cellular thermal shift assay (CETSA) is a label-free technique for intracellular target engagement. ${ }^{5}$ It exploits compound-induced stabilization of protein melting temperatures, but CETSA cannot be applied to all targets. ${ }^{6}$ Activity-based protein profiling (ABPP) methods utilize non-selective irreversible covalent ligands, or intracellular fluorescence polarization probes, combined with intracellular reversible ligand competition, to study target engagement. ${ }^{7}$ Proteolysis targeting chimeras (PROTACs) ${ }^{8,15}$ and specific and nongenetic IAP-dependent protein erasers (SNIPERs) ${ }^{9}$ are heterobifunctional molecules that induce rapid and selective protein degradation, via the proteasome, within living cells. One portion of the molecule engages the target protein while the other, attached via a flexible linker, recruits an E3 ligase to ubiquitinate the target, marking it for degradation as part of the cullin-RING finger machinery. ${ }^{10,15}$

We recently reported the development of a high affinity pirin chemical probe 1 (CCT251236, SPR $K_{\mathrm{D}}=44 \mathrm{nM}$ ) discovered through a cell-based phenotypic screen for inhibitors of the heat shock transcription factor 1 (HSF1) stress pathway. ${ }^{11}$ Pirin is an iron-binding member of the cupin super family of proteins and has been reported as a putative transcription factor regulator ${ }^{12,13}$ It has no known enzymatic function in mammalian cells, no endogenous ligands have been reported and no validated proximal biomarkers have been described. ${ }^{14}$ This makes demonstrating intracellular target engagement in living cells very challenging.

We hypothesized that we could demonstrate chemical probe $\mathbf{1}$ binding to pirin within living cells by developing a pirin-targeting Protein Degradation Probe (PDP).

## RESULTS AND DISCUSSION

Protein Degradation Probe Design. PDPs have been described against various proteins, ${ }^{15}$ with the bromodomain epigenetic target, BRD4, the most extensively studied. ${ }^{16}$ These heterobifunctional PDP molecules have utilized several ligands that bind to the E3 ligases: VHL, ${ }^{17}$ the IAP proteins, ${ }^{18}$ and CRBN. ${ }^{19}$ Despite the rapid expansion in PDP research, there remains no clear methodology to determine which proteins are amenable to PDP-mediated degradation, ${ }^{20}$ which E3 ligase ligand should be exploited or the optimal features of probe design. The structure of the linker, its length and physicochemical properties, have all been demonstrated to be important for PDP activity. ${ }^{21}$ The linker controls the formation of the essential ternary complex, ${ }^{22}$ with evidence that it may stabilize the protein-protein interaction (PPI) between the target and the E3 ligase, rather than forming a detached linear ternary complex (figure 1), although it is unclear whether the ternary complex is part of the multi-protein cullin-RING finger complex. ${ }^{23,2425}$


Figure 1. (Top) Pirin (5JCT)/CRBN (5CI1)/PDP ternary complex design model. The PDP can either stabilize a PPI or simply bring the proteins in close proximity, depending on the role of the linker. (Bottom Left) Cartoon representation of the chemical probe $\mathbf{1}$ (yellow) bound to recombinant pirin (5JCT). The cloud represents the shape of the binding pocket with key residues shown in black and the metal in orange. Red=oxygen, blue=nitrogen. Hydrogens and solvent are omitted for clarity except the water coordinated to the metal, shown as a red sphere. Both representations were generated using the PyMOL Molecular Graphics System, Version 1.8 Schrödinger, LLC. (Bottom Right) Key residues in the binding site and the clear solvent exposed vector for the chemical probe $\mathbf{1}$ binding to pirin are shown, adapted from an analysis using MOE 2014.09. The ethyl pyrrolidine solubilizing group of chemical probe $\mathbf{1}$ was not resolved in the crystal structure and therefore is not drawn in the analysis.

Although we had discovered a high affinity ligand for pirin, there was no evidence that this protein would be a suitable substrate for ubiquitination by a PDP-recruited E3 ligase, without which, extensive linker optimization could be futile. Therefore, we initially designed a synthetically tractable 15 -atom linker that we predicted would not affect the affinity of the PDP for the isolated target proteins. ${ }^{26}$ Analysis of the crystal structure of the chemical probe $\mathbf{1}$ bound
to pirin (figure 1, PDB: 5JCT) ${ }^{11}$ suggested that the solvent-exposed solubilizing group vector should be amenable to linker attachment. We selected a CRBN-targeting thalidomide ligand as the basis of our E3 ligase binding motif, due to its low molecular weight. The CRBN-targeting ligand would be attached via the solvent exposed hydroxyl group of the 4-hydroxythalidomide analogue 2. ${ }^{27}$ The linker would then attach the pirin- and CRBN-targeting specific binding groups via two amide moieties, to give our first generation pirin-targeting PDP 3 (scheme 1).

Scheme 1. Design and Synthesis of the First Generation PDP 3 based on the Chemical
Probe 1


Reagents and conditions: (a) i) $\mathrm{PPh}_{3}$, ${ }^{\mathrm{t}}$ Butyl 2-hydroxyacetate, DTBAD, THF, $0^{\circ} \mathrm{C} \rightarrow \mathrm{RT}, 16 \mathrm{~h}$, $75 \%$; ii) $\mathrm{HCO}_{2} \mathrm{H}, \mathrm{DCM}, 40{ }^{\circ} \mathrm{C}, 16 \mathrm{~h}, 54 \%$; iii) HATU, DIPEA, DMF, ${ }^{\mathrm{t}}$ Butyl (2-(2aminoethoxy)ethyl)carbamate, RT, $16 \mathrm{~h}, 81 \%$; iv) 4 M HCl in dioxane, $0.5 \mathrm{~h}, 100 \%$; (b) i) ethyl 4-bromobutanoate, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, RT, $16 \mathrm{~h}, 37 \%$; (c) Herrmann's palladacycle, ${ }^{29}{ }^{\mathrm{t}} \mathrm{Bu}_{3} \mathrm{PHBF}_{4}$, $\mathrm{MoCO}_{6}$, DBU, 2-(trimethylsilyl)ethan-1-ol, $130{ }^{\circ} \mathrm{C}, 62 \%$ (d) i) TBAF, THF, RT, 16 h ; (e) i) HATU, DIPEA, DMF, RT, 38\% (over 2 steps); ii) LiOH.H2O, MeOH/THF/H2O, RT, 48 h, 18\% (f) HATU, DIPEA, DMF, 74\%. For the synthesis of hydroxythalidomide 2 see reference 28.

The CRBN-targeting motif $\mathbf{4}$ of PDP $\mathbf{3}$ was synthesized from 4-hydroxythalidomide $\mathbf{2}$ in 2 steps and $84 \%$ yield, in a similar manner to that previously described. ${ }^{28}$ The pirin-binding motif of the PDP 3 was synthesized from 6-bromoquinoline 5, via a palladium-mediated carbonylation reaction on the ether derivative 6. Trapping the carbonylation intermediate with 2 -(trimethylsilyl)ethan-1-ol gave the silylethyl-protected ester 7, ${ }^{29}$ which facilitated TBAF-
selective hydrolysis. Amide coupling to the previously described bis-aniline derivative $\mathbf{8},{ }^{11}$ was followed by hydrolysis of the aliphatic linker ester to give acid 9 . Final amide coupling to the CRBN-targeting derivative $\mathbf{4}$ gave the first generation pirin-targeting PDP $\mathbf{3}$ in 10 steps and $0.4 \%$ overall yield.

The First Generation PDP. Analysis of the heterobifunctional PDP 3 revealed that it possessed good affinity for recombinant pirin (table 1, entry 1) when measured using SPR, confirming the success of the rationally designed attachment vector. ${ }^{30}$ The affinity of PDP $\mathbf{3}$ was then assessed against the CRBN-DDB1 complex, with DDB1 acting as a scaffolding protein, using an FPassay similar to that previously described (figure S6). ${ }^{31}$ PDP 3 displayed moderate affinity for CRBN-DDB1, with $K_{\mathrm{i}}=230 \mathrm{nM},{ }^{32}$ comparable to the affinities of the parent CRBN ligands, thalidomide and lenalidomide (figure S7).

Following confirmation that both binding motifs of the PDP 3 retained high affinity for their respective targets, we then investigated its activity against pirin in human cancer cell lines. Several cell lines were assessed for CRBN expression by quantitative capillary electrophoresis (figure S17). ${ }^{33}$ The SK-OV-3 ovarian carcinoma cell line ${ }^{34}$ displayed good basal CRBN and also pirin expression and there was no observable depletion of pirin in these cells when treated with chemical probe 1 ( $1 \mu \mathrm{M}$, data not shown), so this line was selected for further study. Unfortunately, treatment of SK-OV-3 cells with PDP 3, at high concentrations ( $>1 \mu \mathrm{M}$ ) and for extended time periods ( $>48 \mathrm{~h}$ ), resulted in no measurable effects on the cancer cells (data not shown).

The Second Generation PDP. We speculated several causes for the failure of the first generation PDP 3. Pirin may simply be incompatible with this methodology, and if so, no further improvements could be made. ${ }^{35}$ Alternatively, CRBN could be the wrong E3 ligase target to
deplete pirin ${ }^{36}$ or the linker length could be inconsistent with formation of the ternary complex. ${ }^{26}$
Finally, the physicochemical properties of PDP $\mathbf{3}$ could be limiting its intracellular free concentration. ${ }^{37}$ Given these variables, designing a PDP against a protein target that has not previously been validated as being susceptible to E3 ligase-directed degradation is challenging, as optimization cycles can appear lengthy and the difficult and low yielding synthesis discourages the generation of multiple analogues. It was also unclear how strict and narrow the requirements for optimal PDP design would be. We hypothesized that the physicochemical properties of the PDP were the primary cause of the failure of the first generation probe, so began a redesign process that should increase cell membrane flux, whilst maintaining the same linker length and CRBN-targeting ligand (Table 1). By carrying out multiple changes to the PDP, we aimed to minimize the number of iterative design cycles, so we could more rapidly validate this approach.

Table 1. Physicochemical Properties and Affinities for Recombinant Protein Targets of the Three Generations of PDPs

| Entry |  |  |  |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Compd | X | R | Linker | $\mathrm{HBD}^{\text {a }}$ | ALogP ${ }^{\text {b }}$ | $\operatorname{LogD}_{7.4}{ }^{\text {c }}$ | $\begin{aligned} & \text { tPSA } \\ & \left(\AA^{2}\right)^{\mathrm{d}} \end{aligned}$ | $\begin{aligned} & \mathrm{KS} \\ & (\mu \mathrm{M})^{\mathrm{e}} \end{aligned}$ | $\begin{gathered} \text { Pirin } \\ \mathrm{SPR} / K_{\mathrm{D}} \\ \left(\mathrm{p} K_{\mathrm{D}} \pm \mathrm{SEM}\right)^{\mathrm{f}} \end{gathered}$ | $\begin{gathered} \text { CRBN/DDB1 } \\ \text { FP/IC } \\ \left(\mathrm{pIC}_{50} \pm \mathrm{SEM}\right)^{g} \\ \hline \end{gathered}$ | $\underset{(\mathrm{nM})^{\mathrm{h}}}{\mathrm{CRBN} / K_{\mathrm{i}}}$ |
| 1 | 3 | Me |  | $\sim^{0}$ | 5 | 2.8 | 2.2 | 258 | 1 | $\begin{gathered} \hline 110 \mathrm{nM} \\ (6.97 \pm 0.02) \end{gathered}$ | $\begin{gathered} 850 \mathrm{nM} \\ (6.07 \pm 0.14) \end{gathered}$ | 220 |
| 2 | 10 | F |  | 0 | 4 | 2.0 | 1.9 | 244 | 5 | $\begin{gathered} 230 \mathrm{nM} \\ (6.63 \pm 0.01) \end{gathered}$ | $\begin{gathered} 410 \mathrm{nM} \\ (6.38 \pm 0.07) \end{gathered}$ | 95 |
| 3 | 16 | Cl |  | - | 3 | 3.9 | 2.7 | 207 | 2 | $\begin{gathered} 55 \mathrm{nM} \\ (7.26 \pm 0.04) \end{gathered}$ | $\begin{gathered} 490 \mathrm{nM} \\ (6.31 \pm 0.10) \end{gathered}$ | 120 |
| 4 | 21 | Cl |  | 人 | 3 | 3.9 | 2.9 | 207 | 2 | $\underset{(<5.90)}{>1300 \mathrm{nM}}$ | $\begin{gathered} 420 \mathrm{nM} \\ (6.38 \pm 0.10) \end{gathered}$ | 98 |

$\mathrm{SF}=$ significant figure. All data was reprocessed using GraphPad Prism 7.01. See figures S8S16. $\mathrm{SEM}=$ standard error of the mean. ${ }^{\mathrm{a}} \mathrm{HBD}=$ hydrogen bond donor count. ${ }^{\mathrm{b}} \mathrm{ALogP}$ was calculated using Biovia Pipeline Pilot Version 9.5, 2 SF. ${ }^{c} \operatorname{LogD}_{7.4}$ measured using a HPLC-based method, $\mathrm{n}=1,2 \mathrm{SF}$. ${ }^{\mathrm{d}}$ tPSA was calculated using ChemDraw (16.0.1.4) based on the $\mathrm{O}-$ and N count, 3 SF . ${ }^{\mathrm{e}} \mathrm{KS}=$ kinetic solubility in pH 7.4 phosphate buffer at room temperature, $\mathrm{n}=1,1 \mathrm{SF}$. ${ }^{\mathrm{f}} K_{\mathrm{D}}$ values are reported to 2 SF and are calculated by equilibrium analysis using a one site
specific binding model from $\operatorname{SPR}$ sensorgrams at equilibrium where possible, $\mathrm{p} K_{\mathrm{D}}=-\log \left(K_{\mathrm{D}}(\mathrm{M})\right.$ $\mathrm{x} 10^{-9}$ ) and represents the geometric mean of $\mathrm{n}=3$ independent biological repeats. ${ }^{g} \mathrm{IC}_{50}$ values are reported to 2 SF and are calculated from an FP-assay dose-response curve to displace a thalidomide derived fluorescent probe using a $\log [$ Inhibitor $]$ vs. response - Variable slope (four parameters) model, $\mathrm{pIC}_{50}=-\log \left(\mathrm{IC}_{50}(\mathrm{M}) \times 10^{-9}\right)$ and represents the geometric mean of $\mathrm{n}=3$ independent biological repeats, also see reference $31 .{ }^{\mathrm{h}} K_{\mathrm{i}}$ values are calculated from the geometric mean CRBN-DDB1 complex $\mathrm{IC}_{50}$ and the FP-probe $K_{\mathrm{D}}$ using methods described in reference 32 .

The design of heterobifunctional molecules unavoidably results in high molecular weight compounds, making it difficult to balance their physicochemical properties in a manner consistent with acceptable permeability and solubility. ${ }^{38}$ In the case of the first generation pirintargeting PDP 3, although the $\log D_{7.4}$ was acceptable (table 1 , entry 1 ), ${ }^{39}$ the linker had introduced two new hydrogen bond donors (HBD), which can have a negative impact on permeability. The calculated tPSA was very high $\left(258 \AA^{2}\right)$, although it is inevitable that tPSA will be high for large molecules and outside the standard cut-offs for cell permeability, unless the compounds are highly lipophilic. ${ }^{40}$ In design a second generation PDP, we followed standard medicinal chemistry principles to reduce the tPSA and HBD count, whilst maintaining an acceptable $\log \mathrm{D}_{7.4}$. To achieve this, we redesigned the ether linker in the first generation PDP 3 to include a methylene piperazine that would project into solvent, based on analysis of the crystal structure of the chemical probe 1 bound to pirin. The resulting tertiary amide would remove one HBD. We also sought to mask the quinoline amide HBD, using a dipole-dipole interaction, via a bioisosteric replacement with fluorine. ${ }^{41}$ These changes resulted in the design of the second generation pirin-targeting PDP 10 (scheme 2) with a modestly reduced tPSA ( $244 \AA^{2}$ ).

Scheme 2. Synthesis of Second (10) and Third (16) Generation Probes and Control

## Compounds



Reagents and conditions: (a) i) $\mathrm{X}=\mathrm{F}$ 12, 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM; RT, 3 h ; then pyridine, 18 h Quant; ii) $\mathrm{Fe}(0), \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}, 90^{\circ} \mathrm{C}, 1 \mathrm{~h}$, Quant; (b) 2,3-dihydrobenzo-[b][1,4]-dioxine-6-carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine RT, $2 \mathrm{~h}, 85 \%$; (c) i) $\mathrm{SeO}_{2}, 1,4$-dioxane/DMF, reflux, 1 h ; ii) N -Boc-piperazine, DCM, RT, 12 h then $\mathrm{NaBH}(\mathrm{OAc})_{3}$, DCM, RT, $2 \mathrm{~h}, 94 \%$ (over 2 steps); iii) TFA, DCM, RT, 2 h , $69 \%$ (d) i) 2, $\mathrm{PPh}_{3}$, ${ }^{\mathrm{t}}$ Butyl 2-hydroxyacetate, DTBAD, THF, $0{ }^{\circ} \mathrm{C} \rightarrow \mathrm{RT}, 16 \mathrm{~h}, 75 \%$; ii) $\mathrm{HCO}_{2} \mathrm{H}$, DCM, $40^{\circ} \mathrm{C}, 16 \mathrm{~h}, 54 \%$; iii) HATU, DIPEA, DMF, 'Butyl 3-(2-aminoethoxy)propanoate, RT, 16 h, $72 \%$; iv) $\mathrm{HCO}_{2} \mathrm{H}, \mathrm{DCM}, 40{ }^{\circ} \mathrm{C}, 6 \mathrm{~h}, 93 \%$ (e) HATU, DIPEA, DMF, RT, $16 \mathrm{~h}, 52 \%$; (f) i) $\mathrm{X}=\mathrm{Cl} 17$, 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine $2 \mathrm{~h}, 88 \%$; ii) $\mathrm{Fe}(0), \mathrm{NH}_{4} \mathrm{Cl}, \mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O}, 9{ }^{\circ} \mathrm{C}, 1 \mathrm{~h}$, Quant; (g) i) 2,3-dihydrobenzo-[b][1,4]-dioxine-6-carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine 2 h , $61 \%$; (h) i) $\mathrm{SeO}_{2}$, DMF, 1,4-dioxane, $50{ }^{\circ} \mathrm{C}, 16 \mathrm{~h}$; ii) N -Boc-piperazine, $\mathrm{NaBH}_{3} \mathrm{CN}, \mathrm{AcOH}$, DMF, $0^{\circ} \mathrm{C} \rightarrow \mathrm{RT}$, 16 h iii) 4 M HCl in dioxane, MeOH $0{ }^{\circ} \mathrm{C} \rightarrow \mathrm{RT}$, $16 \mathrm{~h}, 32 \%$ over 3 steps; (i) 2-[2-(2-hydroxyethoxy)ethoxy]ethyl 4-methylbenzenesulfonate, $\mathrm{K}_{2} \mathrm{CO}_{3}$, DMF, RT, $16 \mathrm{~h}, 48 \% ; \mathrm{j}$ ) $\mathrm{PPh}_{3}$, DTBAD, THF, RT, $2 \mathrm{~h}, 27 \%$; (k) i) 23, 2,3-dihydro-1,4-benzodioxine-5-carboxylic acid, oxalyl chloride, DMF, DCM, RT, 2 h ; then pyridine $48 \mathrm{~h}, 84 \%$; ii) $\mathrm{Pd} / \mathrm{C}, \mathrm{EtOH} / \mathrm{DCM}, \mathrm{H}_{2}$ ( 1 atm), $77 \%$; iii) 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM; RT, 2 h ; then pyridine $16 \mathrm{~h}, 58 \%$; then same procedure as from $\mathbf{1 8}, 6 \%$ yield over 5 steps. (l) $\mathrm{SeO}_{2}, 1,4-$ dioxane/DMF, $50^{\circ} \mathrm{C}, 5.5 \mathrm{~h}$; ii) N -ethylpiperazine, DCM, RT, 20 h ; then $\mathrm{NaBH}(\mathrm{OAc})_{3}$, DCM, RT, 2 h, $35 \%$ (over 2 steps).

The synthesis of PDP 10 began from the fluoroaniline carboxamide 11, synthesized in a similar manner to that previously described from 2-fluoro-5-nitroaniline $\mathbf{1 2}$ in 3 steps and $85 \%$ yield. Amide coupling to give bisamide $\mathbf{1 3}$ and benzylic oxidation using selenium dioxide, ${ }^{42}$ was followed by reductive amination of the resulting aldehyde with $N$-Boc-piperazine, and subsequent deprotection, to give 14. The CRBN-targeting thalidomide derivative precursor $\mathbf{1 5}$ was prepared in 4 steps and 27\% yield from 4-hydroxythalidomide $\mathbf{2}$. Final amide coupling with monosubstituted piperazine $\mathbf{1 4}$ gave the second generation pirin-targeting PDP 10 in 11 steps and 8\% overall yield (scheme 2).

PDP 10 displayed a similar affinity for recombinant pirin and CRBN-DDB1 to our first generation probe 3 (table 1, entry 2), so was progressed to cellular assessment of pirin degradation. Pleasingly, treatment of SK-OV-3 ovarian cancer cells with $\mathbf{1 0}$ at $3.0 \mu \mathrm{M}$ total concentration for up to 48 h revealed a clear and time-dependent depletion of intracellular pirin expression (figure S18). ${ }^{43}$ This confirmed, for the first time, that pirin is amenable to modulation using a PDP and that the bisamide chemotype not only binds recombinant pirin with high affinity, but also binds pirin within living cells.

Although demonstrating pirin depletion with the second generation probe $\mathbf{1 0}$ was a very encouraging result, we found the effects were poorly reproducible. The concentrations of PDP $\mathbf{1 0}$ needed to observe pirin degradation were high and close to its kinetic solubility (KS). Furthermore, at least 24 h of compound exposure was needed before pirin degradation was observed (figure S18). To explore the cause of the variable results obtained with PDP 10, we assessed its chemical stability. At room temperature, PDP 10 was stable as a solid and in DMSO stock solution ( $>1$ month, data not shown), but at $37{ }^{\circ} \mathrm{C}$ in pH 7.4 phosphate buffer, consistent with the cell assay conditions, it underwent rapid decomposition (table S1), displaying a half-life
of only $\sim 4 \mathrm{~h}$. This poor chemical stability was consistent with the known decomposition of the parent CRBN-targeting thalidomide ligand under these conditions, where multiple hydrolysis products of the imide and glutaramide moieties are observed. ${ }^{44}$ The chemical probe $\mathbf{1}$ displayed no instability under these conditions; therefore, we concluded that the facile hydrolysis of the CRBN-targeting motif limited the reproducibility of our slow-acting second generation pirintargeting PDP 10.

The Third Generation PDP. In the design of a third generation probe, we aimed to increase permeability further. This should result in higher intracellular free concentrations more quickly, mitigating the poor stability of the CRBN-targeting motif. We decided to carry out a bioisosteric replacement of the central ring fluorine for the larger and more sterically hindering chlorine substituent. ${ }^{45}$ However, we were concerned that the increase in lipophilicity from this exchange would negatively impact both the solubility and permeability of the probe. Large lipophilic heterobifunctional molecules are particularly susceptible to aggregation, ${ }^{46}$ and this decreases the free concentration that drives cell membrane flux. ${ }^{47}$ To balance the lipophilicity, we removed the tertiary amide bond to the piperazine, introducing a cationic amine, which would be substantially charged at pH 7.4 (Moka Version 2.5.2, $\mathrm{p} K_{\mathrm{a}}=8.0$ ). ${ }^{48}$ The second amide in the linker was also removed, reducing the HBD count further and increasing the overall flexibility of the ligand, consistent with the formation of the crucial PDP ternary protein complex. The linker length was reduced by one atom to accommodate these changes and resulted in the design of the third generation pirin-targeting PDP 16 (CCT367766), which now displayed a notably reduced, but still high, $\operatorname{tPSA}\left(207 \AA^{2}\right)$.

The synthesis of the pirin-targeting motif of the third generation PDP 16 was carried out in a similar manner to that previously described starting from 2-chloro-5-nitroaniline 17, to give
chlorobisamide 18 in 3 steps and $54 \%$ yield (scheme 2). Oxidation of the methylquinoline moiety with $\mathrm{SeO}_{2}$, reductive amination of the resulting aldehyde with N -Boc-piperazine and N Boc deprotection gave 19 in $32 \%$ yield. Following $\mathrm{S}_{\mathrm{N}} 2$ alkylation with the ether linker to give 20, selective Mitsunobu alkylation with 4-hydroxythalidomide 2 gave the third generation PDP 16 in 8 steps and $2 \%$ overall yield. A non-pirin-binding negative control matched pair compound, PDP 21, based on our negative control pirin chemical probe, ${ }^{11}$ was synthesized from regioisomer $\mathbf{2 3}$ in a similar manner in $2 \%$ overall yield. A non-CRBN binding control $\mathbf{2 2},{ }^{49}$ was also synthesized from 18, utilizing reductive amination with $N$-ethylpiperazine, in $35 \%$ yield (scheme 2).


Figure 2. (A) Representative SPR sensorgram of the third generation PDP 16 and recombinant pirin. (B) Representative binding curve of PDP 16 in the CRBN-DDB1 FP-assay.

Analysis of the third generation PDP 16 confirmed that it retained acceptable lipophilicity (table 1, entry 3), but also displayed a 4.2-fold increase in affinity for recombinant pirin compared to the second generation PDP 10 (figure 2A) and comparable affinity for CRBN (figure 2B). SK-OV-3 cells were then treated with PDP 16 at concentrations from 50 to 1500 nM for up to 24 h (figure 3).


Figure 3. A: Immunoblot of SK-OV-3 human ovarian cancer cells demonstrating the depletion of pirin protein using the third generation PDP 16 and the time-dependent hook-effect. B: Immunoblot demonstrating the concentration-dependent depletion of pirin protein after 2 h exposure in SK-OV-3 cells. C: Capillary electrophoresis and immunoassay were used to quantify the pirin protein expression after 2 h exposure with PDP 16, all values are normalized to vinculin loading control and relative to the measured basal pirin protein expression, all bars represent the arithmetic mean of $\mathrm{n}=3$ independent biological repeats, error bars are SEM. D: Proteomics analysis of the third generation PDP $16(50 \mathrm{nM})$ exposure $(4 \mathrm{~h})$ in SK-OV-3 cells compared to vehicle control, using a tandem mass tagging (TMT) MS2 protocol on the cell lysate, 8547 quantifiable proteins were identified, each blue dot represents a single quantifiable protein, pirin is marked in red (adjusted p value $=1.4 \times 10^{-4}$ ), p values were calculated using a linear modelling based t -test and corrected for multiple comparisons using the Benjamini-Hochberg method to give the $\mathrm{p}(\mathrm{adj})$ values shown, dotted lines represent 2 -fold depletion of the protein and a $\mathrm{p}(\mathrm{adj})=0.05$.

In contrast to our earlier generation PDPs, near complete pirin degradation was now observed with just 50 nM treatment and only 2 h exposure (figure 3 A ). Increasing the total initial concentration of pirin-targeting PDP 16 resulted in a clear hook-effect. This bell-shaped concentration-response is consistent with the formation of a ternary complex. ${ }^{22}$ Interestingly the hook-effect was seen to decrease over time ( 24 h ), possibly either due to the slower degradation at high concentration or from the depletion of PDP 16 due to thalidomide hydrolysis (halflife $=\sim 3 \mathrm{~h}$, figure S 1 ), which would reduce the effective free concentration below the negative cooperativity threshold of the ternary complex. ${ }^{22}$ Pirin degradation was subsequently shown to be concentration-responsive with activity at concentrations as low as 0.5 nM (figure 3B), which was confirmed with a quantitative capillary electrophoreses-based immunoassay (figure 3C). The negative control benzodioxane regioisomer PDP 21, (table 1, entry 4) displayed no pirin depletion at equimolar concentrations (figure S 19 ), ${ }^{50}$ and degradation was also confirmed to be proteasome-dependent by rescue following pre-incubation with the proteosome inhibitor, MG132 (500 nM, figure S19). ${ }^{51}$ We then carried out whole proteome mass spectrometry, to estimate the cellular selectivity of the pirin-targeting PDP 16 in an unbiased manner, quantified using tandem mass tagging (TMT) (figure 3D, www.proteomics.com). After treating SK-OV-3 cells with 50 nM PDP 16 for 4 h , and comparing to vehicle treated cells using BenjaminiHochberg corrected p values, we found that from 8547 quantifiable proteins identified, only pirin (2.3-fold reduction, $\left.\mathrm{p}(\mathrm{adj})=1.4 \times 10^{-4}\right)$ displayed a statistically significant $(\mathrm{p}(\mathrm{adj})<0.05)^{52}$ difference in protein expression.

To confirm that our chemical probe 1 also bound pirin within SK-OV-3 cancer cells, we carried out competition experiments, designed to rescue pirin depletion by PDP 16. The concentrations required for a mutually exclusive binding ligand to displace a probe molecule, are
dependent on the affinity of the ligand relative to the ratio of the free concentration of the probe and its affinity for the protein target complex. ${ }^{53}$ Because the depletion of pirin is a nonequilibrium event that does not necessarily require complete target occupancy, it is more difficult to observe competition at later time points owing to continued protein turnover. ${ }^{54}$ Pretreating SK-OV-3 cells with $10 \mu \mathrm{M}$ thalidomide, as a CRBN-binding competitive ligand, demonstrated that after 2 h treatment with PDP $16(5 \mathrm{nM})$ we successfully rescued pirin depletion (figure S19). The non-CRBN-binding control chlorobisamide 22, displayed high affinity for recombinant pirin (SPR, $K_{\mathrm{D}}=21 \mathrm{nM}, \mathrm{p} K_{\mathrm{D}}=7.68 \pm 0.03, \mathrm{n}=3$ ). Pretreatment of SK-OV-3 cells with chlorobisamide 22 displayed concentration-dependent rescue of pirin expression, following treatment with the PDP 16 (figure 4, top), with complete rescue observed at $1 \mu \mathrm{M}$. Finally, pre-treatment with the chemical probe 1 demonstrated clear rescue of pirin depletion, confirming intracellular target engagement (figure 4, bottom).


Figure 4. Intracellular competition studies with PDP 16 and the chemical probes. SK-OV-3 cells were pre-treated with increasing concentrations of chemical probe for 4 h before exposing to

PDP 16 for 2 h at the concentrations shown. Cells were then lysed and protein expression analyzed using immunoblot. For clarity, gel images have been cropped where appropriate.

## CONCLUSION

Exploiting cell-based phenotypic screens to identify new disease-associated therapeutic targets is an increasingly frequent strategy in drug discovery. While this approach can identify novel targets with unique mechanisms of action, these proteins are often poorly characterized and can lack identifiable enzymatic activity, ligands and biomarkers of target engagement. The main focus of research into PDPs has been as potential therapeutics with novel mechanisms of action. We designed a PDP as an intracellular probe against the poorly-understood and non-catalytic molecular target, pirin. Developing PDPs to confirm intracellular target engagement, and potentially develop intracellular SAR, against challenging proteins, is an important addition to the current methods for compound profiling.

For PDPs to be used as target engagement probes, their rapid development and validation is crucial. The ideal strategies for efficient and successful PDP design are still under investigation and will clearly improve as more protein targets are modulated and additional crystallographic evidence of the target protein/E3 ligase/PDP ternary complexes are discovered. The number of variables involved in PDP design against non-validated target proteins can make the process daunting. By focusing on the physicochemical properties of our probe molecules, in only three iterations, we developed a selective degradation probe that eliminates pirin at low concentration and in a short time-period. This confirmed our chemical probe $\mathbf{1}$ does bind pirin in an intracellular environment and PDP $\mathbf{1 6}$ provides another chemical tool to study a largely unexplored protein.

## EXPERIMENTAL SECTION

## General Experimental.

Unless otherwise stated, reactions were conducted in oven dried glassware under an atmosphere of nitrogen or argon using anhydrous solvents. All commercially obtained reagents and solvents were used as received. Thin layer chromatography (TLC) was performed on precoated aluminum sheets of silica ( 60 F254 nm, Merck) and visualized using short-wave UV light. Flash column chromatography was carried out on Merck silica gel 60 (partial size 40-65 $\mu \mathrm{m})$. Column chromatography was also performed on a Biotage SP1 or Biotage Isolera Four purification system using Biotage Flash silica cartridges (SNAP KP-Sil) or for reverse phase purifications SNAP Ultra C18 cartridges. Ion exchange chromatography was performed using acidic Isolute Flash SCX-II columns. Semi-preparative HPLC was performed on an Agilent 6120 system, flow $20 \mathrm{~mL} / \mathrm{min}$, eluents $0.1 \%$ acetic acid in water and $0.1 \%$ acetic acid in methanol, gradient of $10 \%$ to $100 \%$ organic phase. Lipophilic method: Chromatographic separation at room temperature was carried out using a 1200 Series Preparative HPLC (Agilent, Santa Clara, USA) over a 15 minute gradient elution (Grad15min,20mls) from 60:40 to 0:100 water:methanol (both modified with $0.1 \%$ formic acid) at a flow rate of $20 \mathrm{~mL} / \mathrm{min} .{ }^{1} \mathrm{H}-\mathrm{NMR}$ spectra were recorded on Bruker AMX500 $(500 \mathrm{MHz})$ spectrometers using an internal deuterium lock. Chemical shifts are quoted in parts per million ( ppm ) using the following internal references: $\mathrm{CDCl}_{3}(\delta \mathrm{H} 7.26)$, $\mathrm{MeOD}(\delta \mathrm{H} 3.31)$ and $\mathrm{DMSO}_{\mathrm{d}}(\delta \mathrm{H} 2.50)$. Signal multiplicities are recorded as singlet (s), doublet (d), triplet ( t ), quartet ( q ) and multiplet ( m ), doublet of doublets (dd), doublet of doublet of doublets (ddd), broad (br) or obscured (obs). Coupling constants, $J$, are measured to the nearest $0.1 \mathrm{~Hz} .{ }^{13} \mathrm{C}-\mathrm{NMR}$ spectra were recorded on Bruker AMX500 spectrometers at 126 MHz using an internal deuterium lock. Chemical shifts are quoted to 0.01 ppm , unless greater accuracy was required, using the following internal references: $\mathrm{CDCl}_{3}(\delta \mathrm{C} 77.0)$, $\mathrm{MeOD}(\delta \mathrm{C}$
49.0) and DMSO- $d_{6}$ ( $\delta$ C 39.5). High resolution mass spectra were recorded on an Agilent 1200 series HPLC and diode array detector coupled to a 6210 time of flight mass spectrometer with dual multimode APCI/ESI source or on a Waters Acquity UPLC and diode array detector coupled to a Waters G2 QToF mass spectrometer fitted with a multimode ESI/APCI source. All compounds were $>95 \%$ purity by LCMS analysis unless otherwise stated.

Thalidomide (http://www.sigmaaldrich.com/catalog/product/sigma/t144?lang=en\&region=GB, $\begin{array}{llll}\text { Accessed 2017), } & \text { 29gust } & \text { Lenalidomide }\end{array}$ (http://www.sigmaaldrich.com/catalog/product/aldrich/cds022536?lang=en\&region=GB, $\begin{array}{llll}\text { Accessed 2017) August 29, and } & \text { MG132 }\end{array}$ (http://www.sigmaaldrich.com/catalog/product/sigma/m8699?lang=en\&region=GB, Accessed August 29, 2017) were purchased from Sigma Aldrich and used without further purification.

The pirin chemical probe 1 CCT251236 was synthesized using the previously described procedure. ${ }^{11}$

## 2-(2,6-Dioxopiperidin-3-yl)-4-hydroxyisoindoline-1,3-dione 2

3-Aminopiperidine-2,6-dione hydrochloride ( $300 \mathrm{mg}, 1.82 \mathrm{mmol}$ ) was added to a solution of 4-hydroxyisobenzofuran-1,3-dione ( $299 \mathrm{mg}, 1.82 \mathrm{mmol}$ ) and triethylamine ( $0.38 \mathrm{~mL}, 2.73$ mmol ) in anhydrous THF ( 36.5 mL ) and the reaction heated to reflux for 24 h under inert atmosphere. The reaction mixture was allowed to cool to room temperature, then 1-ethyl-3-(3dimethylaminopropyl)carbodiimide ( $384 \mathrm{mg}, 2.00 \mathrm{mmol}$ ) and DMAP ( $22 \mathrm{mg}, 0.18 \mathrm{mmol}$ ) were added. The reaction was heated to reflux for 24 h . Then, the reaction mixture was allowed to cool and the solvents were removed under reduced pressure. The resulting residue was taken up in methanol and passed sequentially through two ion exchange columns (Isolute SCX-II), eluting with methanol to afford the title compound as an amorphous off-white solid ( $422 \mathrm{mg}, 84 \%$ ). ${ }^{1} \mathrm{H}$

NMR (500 MHz, DMSO- $d_{6}$ ) $\delta 11.18$ (s, 1H), 11.09 (s, 1H), 7.65 (dd, $\left.J=8.4,7.2 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.32$ (dd, $J=7.1,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.25(\mathrm{dd}, J=8.4,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.07(\mathrm{dd}, J=12.8,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.88$ (ddd, $J=17.0,13.9,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.46(\mathrm{~m}, 2 \mathrm{H}), 2.02(\mathrm{dtd}, J=13.0,5.3,2.2 \mathrm{~Hz}, 1 \mathrm{H})$. LCMS ( $\mathrm{ESI}^{+}$) $\mathrm{RT}=0.77 \mathrm{~min}, 100 \%, \mathrm{M}+\mathrm{H}^{+} 275, \mathrm{M}+\mathrm{Na}^{+} 297 .{ }^{28}$ This compound is also commercially available from several suppliers.
tert-Butyl 2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetate
Triphenylphosphine $(0.261 \mathrm{~g}, 0.996 \mathrm{mmol})$ and tert-butyl 2-hydroxyacetate ( $101 \mathrm{mg}, 0.77$ mmol) were dissolved in anhydrous THF ( 3 mL ), while stirring at $0^{\circ} \mathrm{C}$. Then, a solution of DTBAD ( $229 \mathrm{mg}, 0.10 \mathrm{mmol}$ ) in anhydrous THF ( 2 mL ), was added dropwise. Finally, a solution of (2-(2,6-dioxopiperidin-3-yl)-4-hydroxyisoindoline-1,3-dione) $\mathbf{2}$ in anhydrous THF (3 mL ) was added. The mixture was stirred for 1 h at $0^{\circ} \mathrm{C}$, then allowed to warm to room temperature and stirred overnight. Then, the solvent was removed under reduced pressure. The crude product was purified by Biotage chromatography using a gradient of 0 to $50 \% \mathrm{EtOAc}$ in cyclohexane, to afford the title compound as an amorphous white solid ( $224 \mathrm{mg}, 0.577 \mathrm{mmol}, 75$ \% yield). ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 11.11(\mathrm{~s}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~d}$, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{dd}, J=12.8,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.97(\mathrm{~s}, 2 \mathrm{H}), 2.89$ (ddd, $J=17.0,13.9,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.52(\mathrm{~m}, 2 \mathrm{H}), 2.07-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.43(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $d_{6}$ ) $\delta 172.78,169.89,167.14,166.71,165.12,155.03,136.76,133.25,119.96$, $116.44,115.90,81.92,65.50,48.80,30.95,27.68,21.97$. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{15} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{O}_{7}$ $\left(\mathrm{M}+\mathrm{H},-{ }^{\mathrm{t}} \mathrm{Bu}\right)^{+}$333.0717, found 333.0722.

2-((2-(2,6-Dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetic acid
tert-Butyl 2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetate (1.30 g, 3.35 mmol) was dissolved in anhydrous DCM, then formic acid ( $12.8 \mathrm{~mL}, 335 \mathrm{mmol}$ ), was added while stirring at room temperature. The reaction was stirred overnight at $40^{\circ} \mathrm{C}$. The solvents were removed under reduced pressure and the resulting residue purified by reverse phase Biotage chromatography using a gradient of $0 \%$ to $50 \% \mathrm{MeOH}$ in water $+0.1 \%$ formic acid, to afford the title compound as an amorphous white solid ( $600 \mathrm{mg}, 1.81 \mathrm{mmol}, 54 \%$ yield). ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 13.29(\mathrm{~s}, 1 \mathrm{H}), 11.11(\mathrm{~s}, 1 \mathrm{H}), 7.79(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{~d}, J=7.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{dd}, J=12.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~s}, 2 \mathrm{H}), 2.89(\mathrm{ddd}, J=$ $16.9,13.9,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.52(\mathrm{~m}, 2 \mathrm{H}), 2.04(\mathrm{dtd}, J=13.0,5.4,2.3 \mathrm{~Hz}, 1 \mathrm{H}) . \operatorname{LCMS}\left(\mathrm{ESI}^{+}\right):$ $\mathrm{RT}=0.71 \mathrm{~min}, 100 \%, \mathrm{M}+\mathrm{Na}^{+} 355 .{ }^{28}$
tert-Butyl
3-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-
yl)oxy)acetamido)ethoxy)propanoate
2-((2-(2,6-Dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetic acid (217 mg, 0.65 mmol ), tert-butyl 3-(2-aminoethoxy)propanoate ( $124 \mathrm{mg}, 0.65 \mathrm{mmol}$ ) and DIPEA ( 0.34 mL , 1.96 mmol ) were dissolved in anhydrous DMF ( 3.27 mL ) at room temperature under inert atmosphere. HATU ( $230 \mathrm{mg}, 0.98 \mathrm{mmol}$ ) was added and the reaction stirred overnight. The solvents were removed under reduced pressure and the resulting residue was purified by reverse phase Biotage chromatography using a gradient of $10-90 \% \mathrm{MeOH}$ in water $+0.1 \%$ formic acid to afford the title compound as an amorphous white solid ( $236 \mathrm{mg}, 72 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , MeOD) $\delta 7.82(\mathrm{dd}, J=8.4,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{dd}, J=7.3,0.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.46-7.43(\mathrm{~m}, 1 \mathrm{H}), 5.16$ (dd, $J=12.6,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.78(\mathrm{~s}, 2 \mathrm{H}), 3.71(\mathrm{t}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.62-3.56(\mathrm{~m}, 2 \mathrm{H}), 3.52-3.48$ (m, 2H), 2.90 (ddd, $J=18.0,14.4,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.82-2.70(\mathrm{~m}, 2 \mathrm{H}), 2.50(\mathrm{td}, J=6.2,3.8 \mathrm{~Hz}$,

2H), 2.16 (dddd, $J=10.8,7.9,5.1,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.42(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, MeOD) $\delta$ $174.55,172.81,171.33,170.00,168.31,167.59,156.21,138.16,134.96,121.60,119.30,117.89$, 81.75, 70.01, 69.26, 67.75, 50.56, 40.15, 37.22, 32.20, 28.34, 23.66. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{~N}_{3} \mathrm{O}_{9}(\mathrm{M}+\mathrm{H})^{+}$504.1982, found 504.1981.
tert-Butyl
(2-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4yl)oxy)acetamido)ethoxy)ethyl)carbamate

2-((2-(2,6-Dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetic acid ( $0.19 \mathrm{~g}, 0.56 \mathrm{mmol}$ ), tert-butyl (2-(2-aminoethoxy)ethyl)carbamate $(0.11 \mathrm{~g}, 0.56 \mathrm{mmol})$ and DIPEA ( $0.29 \mathrm{ml}, 1.68$ mmol ) were dissolved in anhydrous DMF ( 2.80 mL ) under inert atmosphere at room temperature. HATU (198 mg, 0.84 mmol ) was added and the reaction stirred for 16 h . The reaction mixture was concentrated in vacuo and the resulting residue purified by Biotage chromatography using a gradient of $0-2 \%$ methanol in ethyl acetate. However, this material was not pure, therefore the material was purified by further Biotage chromatography using a gradient of $1-10 \% \mathrm{EtOH}$ in DCM to afford the title compound as an amorphous white solid ( 243 mg , $81 \%) .{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}_{6}$ ) $\delta 11.11(\mathrm{~s}, 1 \mathrm{H}), 8.02(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{dd}, J=8.5$, $7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.49(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.76(\mathrm{t}, J=4.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{dd}$, $J=12.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~s}, 2 \mathrm{H}), 3.44(\mathrm{t}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.38(\mathrm{t}, J=6.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.31(\mathrm{~d}, J=$ $5.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.08(\mathrm{q}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 2.89(\mathrm{ddd}, J=16.6,13.8,5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.65-2.51(\mathrm{~m}$, 2H), $2.08-1.98(\mathrm{~m}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $\left.d_{6}\right) \delta$ 172.73, 169.85, $166.88,166.80,166.72,165.45,155.59,155.00,136.92,133.03,120.34,116.75,116.02,77.63$, 69.02, 68.55, 67.48, 48.80, 38.38, 30.95, 28.22, 21.99. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{24} \mathrm{H}_{30} \mathrm{~N}_{4} \mathrm{O}_{9} \mathrm{Na}$ $(\mathrm{M}+\mathrm{Na})^{+} 541.1911$, found 541.1915.
$N$-(5-(2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-methylphenyl)-2-(4-((2-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetamido)ethoxy)ethyl)amino)-4-oxobutoxy)quinoline-6-carboxamide 3
$N$-(2-(2-Aminoethoxy)ethyl)-2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4yl)oxy)acetamide hydrochloride $4 \quad(21.0 \mathrm{mg}, \quad 0.046 \mathrm{mmol})$, 4 -((6-((5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-methylphenyl)carbamoyl)quinolin-2yl)oxy)butanoic acid ( $25.0 \mathrm{mg}, 0.046 \mathrm{mmol}$ ) and DIPEA ( $0.032 \mathrm{~mL}, 0.184 \mathrm{mmol}$ ) were dissolved in anhydrous DMF ( 0.23 mL ) under inert atmosphere. HATU ( $16.0 \mathrm{mg}, 0.069 \mathrm{mmol}$ ) was added and the reaction left to stir at room temperature for 2 h . The reaction was concentrated in vacuo, then purified by reverse phase Biotage chromatography using a gradient of 20-100\% MeOH in water $+0.1 \%$ formic acid to afford the title compound as a beige solid ( $32 \mathrm{mg}, 74 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 11.09(\mathrm{~s}, 1 \mathrm{H}), 10.07(\mathrm{~s}, 1 \mathrm{H}), 10.06(\mathrm{~s}, 1 \mathrm{H}), 8.56(\mathrm{~d}, J=1.9 \mathrm{~Hz}$, $1 \mathrm{H}), 8.37(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.21(\mathrm{dd}, J=8.8,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{t}, J=5.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.91(\mathrm{t}, J=$ $5.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.87-7.83(\mathrm{~m}, 2 \mathrm{H}), 7.80(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.58(\mathrm{dd}, J=8.3,2.1 \mathrm{~Hz}, 1 \mathrm{H})$, $7.54(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.51(\mathrm{dd}, J=8.4,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.48(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.5$ $\mathrm{Hz}, 1 \mathrm{H}), 7.24(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.08(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.98(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{dd}, J=$ $12.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~s}, 2 \mathrm{H}), 4.42(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.33-4.26(\mathrm{~m}, 4 \mathrm{H}), 3.45(\mathrm{t}, J=5.7 \mathrm{~Hz}$, $2 \mathrm{H}), 3.42(\mathrm{t}, J=5.9 \mathrm{~Hz}, 2 \mathrm{H}), 3.36-3.28(\mathrm{~m}, J=5.5 \mathrm{~Hz}, 2 \mathrm{H}-$ obscured by water peak), $3.24(\mathrm{q}$, $J=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.89$ (ddd, $J=16.3,13.6,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.52(\mathrm{~m}, 2 \mathrm{H}), 2.29(\mathrm{t}, J=7.3 \mathrm{~Hz}$, 2H), $2.23(\mathrm{~s}, 3 \mathrm{H}), 2.11-1.93(\mathrm{~m}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $d_{6}$ ) $\delta$ 172.74, 171.66, 169.87, 166.90, 166.71, 165.44, 164.91, 164.34, 162.79, 154.96, 147.58, 146.33, 142.92, 140.16, $137.28,136.90,136.33,133.01,130.10,130.00,128.75,128.37,128.10,127.68,126.72,123.99$,
$121.19,120.32,118.61,118.11,116.83,116.74,116.65,116.02,113.94,68.91,68.58,67.48$, $65.38,64.39,64.02,48.80,38.44,38.33,31.78,30.94,24.62,21.99,17.48$. HRMS: calcd for $\mathrm{C}_{49} \mathrm{H}_{48} \mathrm{~N}_{7} \mathrm{O}_{13}(\mathrm{M}+\mathrm{H})^{+}, 942.3305$; found 942.3322 .
$N$-(2-(2-Aminoethoxy)ethyl)-2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4yl)oxy)acetamide 4
tert-Butyl (2-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4yl)oxy)acetamido)ethoxy)ethyl)carbamate ( $24 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) was stirred in 4 M HCl in dioxane $(1.16 \mathrm{~mL}, 4.63 \mathrm{mmol})$ at room temperature. After 30 min , the solution was concentrated and used as the hydrochloride salt without further purification. LCMS $\left(\mathrm{ESI}^{+}\right): \mathrm{RT}=0.63 \mathrm{~min}$, $100 \%, \mathrm{M}+\mathrm{H}^{+} 419$.

## Ethyl 4-((6-bromoquinolin-2-yl)oxy)butanoate 6

6-Bromoquinolin-2(1H)-one $5(200 \mathrm{mg}, 0.89 \mathrm{mmol})$ was dissolved in anhydrous DMF (4.5 mL ) at room temperature under nitrogen and potassium carbonate ( $185 \mathrm{mg}, 1.34 \mathrm{mmol}$ ) was added. The reaction mixture was allowed to stir for 1.5 h , then ethyl 4-bromobutanoate ( 0.26 mL , 1.79 mmol ) was added dropwise and the reaction stirred overnight. The reaction mixture was poured into water $(50 \mathrm{ml})$ and the aqueous layer extracted with $\mathrm{DCM}(3 \times 15 \mathrm{ml})$. The combined organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo to afford the crude product as an orange oil. This material was purified by Biotage chromatography using a gradient of $0-100 \%$ EtOAc in cyclohexane to afford the required O-linked isomer as a white solid (111 mg, 37\%) and the undesired N -linked isomer as a yellow oil ( $164 \mathrm{mg}, 55 \%$ ). The regioisomers were distinguishable by NOE spectroscopy. ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.91(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H})$,
$7.88-7.87(\mathrm{br} \mathrm{m}, 1 \mathrm{H}), 7.71-7.69(\mathrm{~m}, 2 \mathrm{H}), 6.92(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.52(\mathrm{t}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H})$, $4.18(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.55(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.19(\mathrm{ddd}, J=13.7,7.4,6.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.28(\mathrm{t}, J$ $=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 173.39,162.35,145.41,137.80,132.82,129.61$, $129.11,126.42,117.29,114.30,65.17,60.59,31.27,24.60,14.39$. HRMS $^{(E S I}{ }^{+}$): calcd for $\mathrm{C}_{15} \mathrm{H}_{17}{ }^{79} \mathrm{BrNO}_{3}(\mathrm{M}+\mathrm{H})^{+}, 339.0419$; found 339.0403.

2-(Trimethylsilyl)ethyl 2-(4-ethoxy-4-oxobutoxy)quinoline-6-carboxylate 7
Ethyl 4-((6-bromoquinolin-2-yl)oxy)butanoate $\mathbf{6} \quad(554 \mathrm{mg}, 1.64 \mathrm{mmol})$, tri-tertbutylphosphonium tetrafluoroborate ( $95 \mathrm{mg}, 0.33 \mathrm{mmol}$ ) and Herrmann's palladacyle ( 77 mg , 0.08 mmol ) were added to a microwave vial and suspended in 2-(trimethylsilyl)ethanol ( 15 ml ). Molybdenum hexacarbonyl ( $865 \mathrm{mg}, 3.28 \mathrm{mmol}$ ) followed by DBU 1.0 M in THF ( 4.91 mL , 4.91 mmol ) were then added and the vial promptly sealed. The reaction was heated to $130^{\circ} \mathrm{C}$ for 1 h in a microwave. The reaction mixture was diluted with DCM and filtered to remove solids. The filtrate was concentrated in vacuo. This residue was diluted with water ( 20 ml ) and extracted with DCM ( $3 \times 20 \mathrm{ml}$ ). The combined organic layer was washed with brine and concentrated under reduced pressure. The resulting crude was purified by Biotage chromatography using a gradient of $0-100 \%$ EtOAc in cyclohexane to give 472 mg product ( $88 \%$ purity by LCMS, $62 \%$ ), as a brown oil. ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.47(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.23(\mathrm{dd}, J=8.7,1.9 \mathrm{~Hz}$, $1 \mathrm{H}), 8.06(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.84(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{t}, J=6.3$ $\mathrm{Hz}, 2 \mathrm{H}), 4.51-4.43(\mathrm{~m}, 2 \mathrm{H}), 4.17(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.55(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.19(\mathrm{p}, J=7.0$, $6.5 \mathrm{~Hz}, 2 \mathrm{H}), 1.27(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.23-1.15(\mathrm{~m}, 2 \mathrm{H}), 0.12(\mathrm{~s}, 9 \mathrm{H}) . \operatorname{LCMS}\left(\mathrm{ESI}^{+}\right): \mathrm{RT}=1.79$ $\min , 88 \%,(\mathrm{M}+\mathrm{H})^{+} 404$.

Ethyl 4-((6-((5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-methylphenyl)carbamoyl)quinolin-2-yl)oxy)butanoate.

To a stirring solution of 2-(trimethylsilyl)ethyl 2-(4-ethoxy-4-oxobutoxy)quinoline-6carboxylate 7 ( $463 \mathrm{mg}, 1.15 \mathrm{mmol}$ ) in anhydrous THF ( 12 mL ) at room temperature under nitrogen was dropwise added TBAF 1.0 M in THF ( $1.72 \mathrm{~mL}, 1.72 \mathrm{mmol}$ ). The reaction was allowed to stir at RT overnight, after which time the starting material had been consumed as indicated by LCMS. The reaction mixture was diluted with water and concentrated in vacuo to remove the THF. The aqueous layer was acidified to $\sim \mathrm{pH} 2$ with 1 M HCl aq. and extracted with DCM ( 3 x 15 ml ). The combined organic layer was washed with brine ( 20 mL ) and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ to afford the crude product as a yellow oil. This material was used directly in the next step without further purification. LCMS (ESI $)$ : $\mathrm{RT}=1.57 \mathrm{~min}, 84 \%,(\mathrm{M}+\mathrm{H})^{+} 304$.

2-(4-Ethoxy-4-oxobutoxy)quinoline-6-carboxylic acid ( $74 \mathrm{mg}, 0.24 \mathrm{mmol}$ ) was dissolved in anhydrous DMF ( 2 ml ) and DIPEA ( $0.12 \mathrm{ml}, 0.67 \mathrm{mmol}$ ) was added, followed by HATU (105 $\mathrm{mg}, 0.28 \mathrm{mmol})$. The reaction was allowed to stir for 5 min and then N -(3-amino-4-methylphenyl)-2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamide ( $63 \mathrm{mg}, 0.22 \mathrm{mmol}$ ) was added. The reaction was stirred at room temperature overnight. The reaction mixture was poured into water and the resulting precipitate collected by filtration. The precipitate was dissolved in $\mathrm{DCM} / \mathrm{MeOH}$ and pre-absorbed onto silica. The crude material was purified by Biotage chromatography using a gradient of $0-5 \% \mathrm{MeOH}$ in DCM and then by preparative HPLC (20 mls , lipophilic method) to afford the title compound as an amorphous beige solid ( $48 \mathrm{mg}, 38 \%$ over two steps). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.30(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.13-8.05(\mathrm{~m}, 3 \mathrm{H})$, $7.93-7.87(\mathrm{~m}, 3 \mathrm{H}), 7.68(\mathrm{dd}, J=8.3,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.44(\mathrm{~d}, J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.38(\mathrm{dd}, J=8.4,2.2$ $\mathrm{Hz}, 1 \mathrm{H}), 7.23$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.97(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{t}, J=$
6.3 Hz, 2H), 4.31 (ddd, $J=11.1,3.7,1.8 \mathrm{~Hz}, 4 \mathrm{H}), 4.18(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.56(\mathrm{t}, J=7.4 \mathrm{~Hz}$, $2 \mathrm{H}), 2.35(\mathrm{~s}, 3 \mathrm{H}), 2.21(\mathrm{p}, J=6.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.28(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 173.37,165.49,165.09,163.48,148.69,146.89,143.67,139.54,136.92,136.11,131.17$, $130.22,128.23,128.05,127.62,127.30,125.17,124.61,120.59,117.63,117.52,116.89,114.86$, 114.58, 65.40, 64.71, 64.34, 60.62, 31.26, 24.58, 17.52, 14.39. HRMS (ESI ${ }^{+}$: calcd for $\mathrm{C}_{32} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+}, 570.2253$; found 570.2222.
4-((6-((5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-
methylphenyl)carbamoyl)quinolin-2-yl)oxy)butanoic acid 9 .
To a stirring solution of ethyl 4-((6-((5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-methylphenyl)carbamoyl)quinolin-2-yl)oxy)butanoate ( $37 \mathrm{mg}, 0.065 \mathrm{mmol}$ ) in THF ( 0.5 mL ) and $\mathrm{MeOH}(0.25 \mathrm{~mL})$ was added a solution of $\mathrm{LiOH} . \mathrm{H}_{2} \mathrm{O}(13.6 \mathrm{mg}, 0.325 \mathrm{mmol})$ in water $(0.25$ $\mathrm{mL})$. After 2 days the reaction mixture was diluted with water ( 30 ml ) and the aqueous layer acidified to $\sim \mathrm{pH} 2$ with 1 M HCl aq. The aqueous layer was extracted with $\mathrm{DCM}(3 \times 15 \mathrm{ml})$. The combined organic layer was washed with 1 M HCl aq. $(20 \mathrm{ml})$, brine $(20 \mathrm{ml})$ and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, then concentrated in vacuo to afford the product as an amorphous pale yellow solid $(6.5 \mathrm{mg}$, $18 \%$ ). This material was used directly in the next reaction without further purification. ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 12.16$ (br s, 1 H ), $10.07(\mathrm{~s}, 2 \mathrm{H}), 8.56(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.38(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 1 \mathrm{H}), 8.22(\mathrm{dd}, J=8.7,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.89-7.83(\mathrm{~m}, 2 \mathrm{H}), 7.58(\mathrm{dd}, J=8.3,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.54$ (d, $J=2.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.51(\mathrm{dd}, J=8.4,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.24(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.11(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.99(\mathrm{~s}, 1 \mathrm{H}), 4.47(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.34-4.21(\mathrm{~m}, 4 \mathrm{H}), 2.43(\mathrm{t}, J=7.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.24(\mathrm{~s}$, $3 \mathrm{H}), 2.03(\mathrm{p}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $d_{6}$ ) $\delta$ 174.11, 164.94, 164.35, 162.78, 147.57, 146.34, 142.93, 140.23, 137.29, 136.33, 130.11, 130.05, 128.76, 128.40, 128.12, 127.68,
$126.75,124.03,121.20,118.61,118.11,116.84,116.66,113.94,65.06,64.39,64.02,30.30$, 24.05, 17.49. $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right)$: calcd for $\mathrm{C}_{30} \mathrm{H}_{28} \mathrm{~N}_{3} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+}$, 542.1922; found 542.1905.

## N -(2-Fluoro-5-nitrophenyl)-2-methylquinoline-6-carboxamide

Oxalyl chloride ( $3.25 \mathrm{~mL}, 38.4 \mathrm{mmol}$ ) was added dropwise to a solution of 2-methylquinoline-6-carboxylic acid ( $6.59 \mathrm{~g}, 35.2 \mathrm{mmol}$ ) and DMF ( $0.0062 \mathrm{~mL}, 0.080 \mathrm{mmol}$ ) in anhydrous DCM ( 80 mL ). The reaction mixture was stirred at room temperature for 3 h and then concentrated under reduced pressure. The residue was dissolved in DCM ( 30 mL ) and concentrated again under reduced pressure. The resulting dry residue was dissolved in pyridine ( 80 mL ) and 2-fluoro-5-nitroaniline $12(5.00 \mathrm{~g}, 32.0 \mathrm{mmol})$ was added in one portion. The reaction mixture was stirred at room temperature for 18 h and then poured onto water $(100 \mathrm{~mL})$. The green precipitate was filtered and washed with water ( $3 \times 20 \mathrm{~mL}$ ), diethyl ether ( $3 \times 20 \mathrm{~mL}$ ) and DCM ( 10 mL ), to afford the title compound as a light green solid, which was carried onto the next step without further purification (10.4 g, Quant.). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}-\mathrm{d}_{6}$ ) $\delta 10.71$ (s, 1H), 8.72 (dd, $J$ $=6.5,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.63(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.43(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.23(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}$, $1 \mathrm{H}), 8.21-8.16(\mathrm{~m}, 1 \mathrm{H}), 8.05(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.65(\mathrm{app} \mathrm{t}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}$ ), $2.71(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $_{6}$ ) $\delta$ 165.53, 161.22, $158.67(\mathrm{~d}, J=258.2$ $\mathrm{Hz}), 148.65,143.72,137.32,130.36,128.88,128.48,128.00,127.08$ (d, $J=13.9 \mathrm{~Hz}), 125.33$, 123.18, $122.14(\mathrm{~d}, J=9.6 \mathrm{~Hz}), 121.25(\mathrm{~d}, J=3.8 \mathrm{~Hz}), 117.19(\mathrm{~d}, J=22.8 \mathrm{~Hz}), 25.07 .{ }^{19} \mathrm{~F}$ NMR (470 MHz, DMSO-d ${ }_{6}$ ) $\delta-110.20$. HRMS ( $\mathrm{ESI}^{+}$): calcd for $\mathrm{C}_{17} \mathrm{H}_{13} \mathrm{FN}_{3} \mathrm{O}_{3}(\mathrm{M}+\mathrm{H})^{+}, 326.0935$; found 326.0931.
$N$-(5-(2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-((4-(3-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)acetamido)ethoxy)propanoyl)piperazin-1-yl)methyl)quinoline-6-carboxamide $\mathbf{1 0}$

3-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-
yl)oxy)acetamido)ethoxy)propanoic acid $15 \quad(15 \mathrm{mg}, \quad 0.033 \mathrm{mmol}), \quad N \quad N$-(5-(2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-(piperazin-1-ylmethyl)quinoline-6-carboxamide $14(18 \mathrm{mg}, 0.033 \mathrm{mmol})$ and HATU ( $19 \mathrm{mg}, 0.050 \mathrm{mmol}$ ) were dissolved in anhydrous DMF ( 0.34 ml ) at room temperature under inert atmosphere. DIPEA ( $23.4 \mu \mathrm{~L}, 0.134 \mathrm{mmol}$ ) was added and the reaction stirred for 16 h . The reaction mixture was concentrated in vacuo and purified by reverse phase Biotage chromatography (10-80\% MeOH in water $+0.1 \%$ formic acid), however the material obtained was not pure. Further purification by Biotage chromatography using a gradient of 2-50\% EtOH in DCM afforded the product as an off-white amorphous solid ( $17 \mathrm{mg}, 52 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 11.12$ (s, 1H), $10.39(\mathrm{~s}, 1 \mathrm{H}), 10.18(\mathrm{~s}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.50(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 8.25(\mathrm{dd}$, $J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.14(\mathrm{dd}, J=7.1,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.08(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.99(\mathrm{t}, J=5.6 \mathrm{~Hz}$, $1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.65(\mathrm{ddd}, J=9.0,4.4,2.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.56-7.46(\mathrm{~m}, 3 \mathrm{H}), 7.39(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\mathrm{dd}, J=10.1,9.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 1 \mathrm{H}), 5.12(\mathrm{dd}, J=12.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~s}, 2 \mathrm{H}), 4.31(\mathrm{td}, J=5.3,3.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.82(\mathrm{~s}, 2 \mathrm{H})$, $3.62(\mathrm{t}, J=6.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.52-3.41(\mathrm{~m}, 6 \mathrm{H}), 3.33(\mathrm{~s}, 2 \mathrm{H}), 2.89(\mathrm{ddd}, J=16.7,13.7,5.4 \mathrm{~Hz}, 1 \mathrm{H})$, $2.62-2.52(\mathrm{~m}, 4 \mathrm{H}), 2.43(\mathrm{~d}, J=21.7 \mathrm{~Hz}, 4 \mathrm{H}), 2.09-1.99(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \mathrm{NMR}\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 171.71,169.89,168.79,167.12,166.76,166.06,165.15,165.08,161.23,154.62,149.43(\mathrm{~d}, J=$ 240.9 Hz ), 149.10, 146.95, 143.59, 137.71, 137.06, 134.97, 133.71, 131.94, 129.99, 127.99, $127.94(\mathrm{~d}, J=9.5 \mathrm{~Hz}), 127.28,126.81,126.42(\mathrm{~d}, J=11.2 \mathrm{~Hz}), 122.32,120.74,119.68,118.21$,
$117.48,117.46(\mathrm{~d}, J=6.5 \mathrm{~Hz}), 117.05(\mathrm{~d}, J=7.0 \mathrm{~Hz}), 116.93,115.30(\mathrm{~d}, J=20.0 \mathrm{~Hz}), 113.74$, $69.20,68.22,67.52,64.74,64.70,64.32,53.39,53.15,49.42,45.80,41.68,39.47,33.60,31.52$, 23.00. $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right)$: calcd for $\mathrm{C}_{50} \mathrm{H}_{48} \mathrm{FN}_{8} \mathrm{O}_{12}(\mathrm{M}+\mathrm{H})^{+}, 971.3370$; found 971.3343 .

N -(5-Amino-2-fluorophenyl)-2-methylquinoline-6-carboxamide 11
To a solution of $N$-(2-fluoro-5-nitrophenyl)-2-methylquinoline-6-carboxamide (10.4 g, 32.0 $\mathrm{mmol})$ in ethanol $(120 \mathrm{~mL})$ and water $(40 \mathrm{~mL})$, ammonium chloride $(12.0 \mathrm{~g}, 224 \mathrm{mmol})$ and iron powder ( $12.5 \mathrm{~g}, 224 \mathrm{mmol}$ ) were added in one portion and the resulting suspension was allowed to stir at $90^{\circ} \mathrm{C}$ for 1 h . The reaction mixture was allowed to cool to room temperature, diluted with $\mathrm{MeOH}(20 \mathrm{~mL})$ and $\mathrm{DCM}(20 \mathrm{~mL})$ and filtered through a pad of Celite. The resulting filtrate was concentrated under vacuum to afford a light brown solid which was re-dissolved in a mixture of $\mathrm{DCM}: \mathrm{MeOH}(9: 1,150 \mathrm{~mL})$ and washed with saturated aqueous $\mathrm{NaHCO}_{3}(150 \mathrm{~mL})$. The organic phase was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$, filtered and concentrated under reduced pressure to afford a yellow solid as crude product, which was taken directly onto the next step without further purification ( 9.46 g , Quant.). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}_{\mathrm{d}}^{6}$ ) $\delta 10.05(\mathrm{~s}, 1 \mathrm{H}), 8.57(\mathrm{~d}, J=$ $1.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.39(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.19(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.01(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H})$, $7.52(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{dd}, J=10.3,8.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.89(\mathrm{dd}, J=6.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.46-$ $6.39(\mathrm{~m}, 1 \mathrm{H}), 5.05(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 2.70(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $d_{6}$ ) $\delta$ 164.93, 160.84, $148.49,147.72(\mathrm{~d}, J=233.9 \mathrm{~Hz}), 145.08(\mathrm{~d}, J=1.9 \mathrm{~Hz}), 137.19,131.19,128.44,128.33,127.95$, $125.50(\mathrm{~d}, J=13.1 \mathrm{~Hz}), 125.34,122.99,115.54(\mathrm{~d}, J=20.6 \mathrm{~Hz}), 111.52,111.39(\mathrm{~d}, J=6.6 \mathrm{~Hz})$, 25.05. ${ }^{19} \mathrm{~F}$ NMR (470 MHz, DMSO-d ${ }_{6}$ ) $\delta$-138.12. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{FN}_{3} \mathrm{O}(\mathrm{M}+$ H) ${ }^{+}$, 296.1194; found 296.1191.
$N$-(5-(2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-methylquinoline-6carboxamide 13

To a suspension of 2,3-dihydrobenzo[b][1,4]dioxine-6-carboxylic acid ( $12.7 \mathrm{~g}, 70.5 \mathrm{mmol}$ ) in anhydrous DCM ( 100 mL ), a catalytic amount of anhydrous DMF ( $6.16 \mu \mathrm{l}, 0.080 \mathrm{mmol}$ ) and oxalyl chloride ( $6.51 \mathrm{~mL}, 77.0 \mathrm{mmol}$ ) were added dropwise and the resulting green solution was allowed to stir at room temperature for 3 h . After which time, the reaction mixture was concentrated in vacuo to afford a dry pale green solid. The solid was dissolved in pyridine (100 mL ) and $N$-(5-amino-2-fluorophenyl)-2-methylquinoline-6-carboxamide ( $9.46 \mathrm{~g}, 32.0 \mathrm{mmol}$ ) was added in one portion. The resulting dark yellow suspension was allowed to stir for 2 h and was then poured onto water $(100 \mathrm{~mL})$. The yellow precipitate was filtered and washed with water ( $3 \times 20 \mathrm{~mL}$ ), diethyl ether ( $3 \times 20 \mathrm{~mL}$ ) and DCM ( 10 mL ), to afford the crude product as a pale yellow solid, which was taken directly onto the next step without purification ( $12.5 \mathrm{~g}, 85 \%$ ). ${ }^{1} \mathrm{H}$ NMR (500 MHz, DMSO-d $)_{6}$ : $\delta 10.38(\mathrm{~s}, 1 \mathrm{H}), 10.19(\mathrm{~s}, 1 \mathrm{H}), 8.62(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.41(\mathrm{~d}, J$ $=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.23(\mathrm{dd}, J=8.8,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.13(\mathrm{dd}, J=7.0,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.03(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 7.68-7.62(\mathrm{~m}, 1 \mathrm{H}), 7.57-7.46(\mathrm{~m}, 3 \mathrm{H}), 7.29(\mathrm{app} \mathrm{t}, J=9.5,1 \mathrm{H}), 6.99(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H})$, $4.34-4.28(\mathrm{~m}, 4 \mathrm{H}), 2.71(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $d_{6}$ ) $\delta 165.09,164.48,160.95$, $151.83(\mathrm{~d}, J=243.4 \mathrm{~Hz}), 148.57,146.45,142.95,137.22,135.42(\mathrm{~d}, J=2.0 \mathrm{~Hz}), 130.87,128.50$, 128.41, 127.94, 127.47, 125.39 (d, $J=9.5 \mathrm{~Hz}), 125.35,123.05,121.25,118.81,118.75$ (d, $J=$ $13.0 \mathrm{~Hz}), 116.89,116.69,115.56(\mathrm{~d}, J=21.2 \mathrm{~Hz}), 64.41,64.03,25.06 .{ }^{19} \mathrm{~F}$ NMR (470 MHz, DMSO- $\mathrm{d}_{6}$ ) $\delta$-126.65. HRMS (ESI+): calcd for $\mathrm{C}_{26} \mathrm{H}_{21} \mathrm{FN}_{3} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+}, 458.1511$; found 458.1499.
$N$-(5-(2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-(piperazin-1-ylmethyl)quinoline-6-carboxamide 14

A solution of $N$-(5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-methylquinoline-6-carboxamide $\mathbf{1 3}(5.00 \mathrm{~g}, 10.9 \mathrm{mmol})$ and selenium dioxide ( $1.33 \mathrm{~g}, 12.0$ $\mathrm{mmol})$ in anhydrous DMF ( 40 mL ) and 1,4-dioxane ( 120 mL ) was heated at reflux for 1 h . Then, the reaction mixture was allowed to cool to room temperature, diluted with $\mathrm{DCM}(20 \mathrm{~mL})$ and filtered through a pad of Celite. The filtrate was concentrated under vacuum (using a heptane/EtOAc azeotrope to remove DMF) to afford the crude product as a yellow solid, which was carried onto the next step without further purification $(5.15 \mathrm{~g})$.

A solution of $N$-(5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)-2-formylquinoline-6-carboxamide ( $255 \mathrm{mg}, 0.541 \mathrm{mmol}$ ) and tert-butyl piperazine-1-carboxylate ( $302 \mathrm{mg}, 1.62 \mathrm{mmol}$ ) in anhydrous $\mathrm{DCM}(5 \mathrm{~mL})$ was allowed to stir at $20^{\circ} \mathrm{C}$ for 12 h , after which time, sodium triacetoxyborohydride ( $344 \mathrm{mg}, 1.62 \mathrm{mmol}$ ) was added in one portion and the resulting mixture was allowed to stir at $20^{\circ} \mathrm{C}$ for 2 h . The reaction was quenched with $\mathrm{NaHCO}_{3}$ saturated aqueous solution $(5 \mathrm{~mL})$ and extracted with a DCM:MeOH, 9:1 mixture (3 x 5 mL ). The crude product (pale yellow solid) was purified by column chromatography on silica gel using a gradient of $0-6 \% \mathrm{MeOH}$ in DCM, followed by trituration in diethyl ether to afford the desired product as a pale beige solid ( $325 \mathrm{mg}, 94 \%$ ).

To a suspension of tert-butyl 4-((6-((5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)-2-fluorophenyl)carbamoyl)quinolin-2-yl)methyl)piperazine-1-carboxylate ( $300 \mathrm{mg}, 0.468 \mathrm{mmol}$ ) in anhydrous DCM ( 5 mL ) , TFA $(0.179 \mathrm{~mL}, 2.33 \mathrm{mmol})$ was added dropwise and the resulting mixture was allowed to stir at $20^{\circ} \mathrm{C}$ for 3 h . The reaction mixture was concentrated under reduced pressure to afford the crude product as a light brown oil. The crude was purified by
column chromatography on silica gel using a gradient of $0-15 \% \mathrm{MeOH}$ in DCM , followed by trituration in diethyl ether to afford the title compound as a white solid ( $174 \mathrm{mg}, 69 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (500 MHz, DMSO- $d_{6}$ ): $\delta 10.42(\mathrm{~s}, 1 \mathrm{H}), 10.20(\mathrm{~s}, 1 \mathrm{H}), 8.73(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 8.67(\mathrm{~d}, J=1.66 \mathrm{~Hz}, 1 \mathrm{H})$, 8.52 (d, $J=8.71 \mathrm{~Hz}, 1 \mathrm{H}), 8.27$ (dd, $J=8.71,1.66 \mathrm{~Hz}, 1 \mathrm{H}), 8.15$ (dd, $J=7.08,2.72 \mathrm{~Hz}, 1 \mathrm{H}), 8.09$ (d, $J=8.71 \mathrm{~Hz}, 1 \mathrm{H}), 7.74(\mathrm{~d}, J=8.42 \mathrm{~Hz}, 1 \mathrm{H}), 7.66-7.61(\mathrm{~m}, 1 \mathrm{H}), 7.54(\mathrm{~d}, J=2.11 \mathrm{~Hz}, 1 \mathrm{H})$, $7.52(\mathrm{dd}, J=8.42,2.11 \mathrm{~Hz}, 1 \mathrm{H}), 7.30(\operatorname{app} \mathrm{t}, J=10.10 \mathrm{~Hz}, 1 \mathrm{H}), 6.99(\mathrm{~d}, J=8.42 \mathrm{~Hz}, 1 \mathrm{H}), 4.35-$ $4.27(\mathrm{~m}, 4 \mathrm{H}), 3.90(\mathrm{~s}, 2 \mathrm{H}), 3.19-3.09(\mathrm{~m}, 4 \mathrm{H}), 2.76-2.66(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $d_{6}$ ) $\delta 164.98,164.50,160.59,151.87(\mathrm{~d}, ~ J=243.8 \mathrm{~Hz}), 148.27,146.47,142.96,137.76$, $135.45(\mathrm{~d}, ~ J=2.7 \mathrm{~Hz}), 131.45,128.79,128.57,128.13,127.46,126.25,125.30(\mathrm{~d}, J=13.2 \mathrm{~Hz})$, 121.95, 121.26, 118.86, 118.78 (d, $J=7.8 \mathrm{~Hz}$ ), 116.90, $116.70,115.59(\mathrm{~d}, J=20.9 \mathrm{~Hz}), 64.42$, 64.04, 63.67, 49.45, 43.03. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{30} \mathrm{H}_{29} \mathrm{FN}_{5} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+}, 542.2198$; found 542.2190 .

## 3-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-

yl)oxy)acetamido)ethoxy)propanoic acid 15
tert-Butyl 3-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4yl)oxy)acetamido)ethoxy)propanoate ( $29 \mathrm{mg}, 0.06 \mathrm{mmol}$ ) and formic acid ( $0.22 \mathrm{ml}, 5.76 \mathrm{mmol}$ ) were dissolved in anhydrous $\mathrm{DCM}(0.29 \mathrm{~mL})$ at $40^{\circ} \mathrm{C}$ for 6 h . The reaction mixture was concentrated in vacuo and the resulting residue purified by reverse phase Biotage chromatography using a gradient of $10-80 \% \mathrm{MeOH}$ in water $+0.1 \%$ formic acid to afford the title compound as an amorphous white solid ( $24 \mathrm{mg}, 93 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta$ $12.21(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 11.12(\mathrm{~s}, 1 \mathrm{H}), 8.01(\mathrm{t}, J=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.81(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.50(\mathrm{~d}, J$ $=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.39(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.11(\mathrm{dd}, J=12.9,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.79(\mathrm{~s}, 2 \mathrm{H}), 3.60(\mathrm{t}, J=$
$6.4 \mathrm{~Hz}, 2 \mathrm{H}), 3.44(\mathrm{t}, J=5.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.30(\mathrm{q}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H}$ - partially obscured by water peak), $2.94-2.82(\mathrm{~m}, 1 \mathrm{H}), 2.64-2.51(\mathrm{~m}, 2 \mathrm{H}), 2.48-2.41(\mathrm{~m}, 2 \mathrm{H}), 2.08-1.99(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $d_{6}$ ) $\delta 172.80,172.69,169.91,166.92,166.77,165.48,155.00,136.98,133.06$, 120.36, 116.78, 116.07, 68.58, 67.49, 66.06, 48.84, 38.32, 34.67, 30.98, 22.01. HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{~N}_{3} \mathrm{O}_{9}(\mathrm{M}+\mathrm{H})^{+} 448.1356$, found 448.1351 .
$N$-(2-Chloro-5-nitrophenyl)-2-methylquinoline-6-carboxamide
2-Methylquinoline-6-carboxylic acid $(1.50 \mathrm{~g}, 8.01 \mathrm{mmol})$ was suspended in anhydrous DCM $(40 \mathrm{~mL})$ under inert atmosphere. DMF ( $1.40 \mu \mathrm{l}, 0.018 \mathrm{mmol}$ ) and oxalyl chloride ( $0.74 \mathrm{~mL}, 8.74$ mmol ) were added dropwise and the resulting green solution was allowed to stir at $20^{\circ} \mathrm{C}$ for 3 h , after which time it was concentrated in vacuo to afford a dry pale green solid. The solid was dissolved in pyridine $(40.0 \mathrm{~mL})$ and 2-chloro-5-nitroaniline $17(1.26 \mathrm{~g}, 7.28 \mathrm{mmol})$ was added in one portion. The resulting dark yellow suspension was allowed to stir for 2 h , then it was poured onto water and the yellow precipitate was filtered and washed several times with water, $\mathrm{Et}_{2} \mathrm{O}$ and finally with a minimum amount of DCM to afford the crude product as a yellow amorphous solid which was used without further purification ( $2.20 \mathrm{~g}, 88 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}\right.$, DMSO- $\left.d_{6}\right): \delta$ $10.59(\mathrm{~s}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.60(\mathrm{~d}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.44(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.25$ (dd, $J=8.6,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.15(\mathrm{dd}, J=8.6,2.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.06(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.91(\mathrm{~d}, J=$ $8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.55(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.71(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{HRMS}_{\left(\mathrm{ESI}^{+}\right): \text {Found }[\mathrm{M}+\mathrm{H}]^{+} 342.0646}$ $\mathrm{C}_{17} \mathrm{H}_{13}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{3}$ requires 342.0640 .
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-((4-(2-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)ethoxy)ethoxy)ethyl)piperazin-1-yl)methyl)quinoline-6-carboxamide 16 (CCT367766)

2-(2,6-Dioxo-3-piperidyl)-4-hydroxy-isoindoline-1,3-dione 2 ( $29 \mathrm{mg}, 0.100 \mathrm{mmol}$ ) was dissolved in anhydrous THF ( 1 ml ) under argon and then triphenylphosphine ( $29 \mathrm{mg}, 0.110$ mmol) was added and $N$-(2-chloro-5-(2,3-dihydro-1,4-benzodioxine-6-carbonylamino)phenyl)-2-((4-(2-(2-(2-hydroxyethoxy)ethoxy)ethyl)piperazin-1-yl)methyl)quinoline-6-carboxamide 20 $(72.00 \mathrm{mg}, \quad 0.1000 \mathrm{mmol})$ were added, followed by tert-butyl (NE)-N-tertbutoxycarbonyliminocarbamate $(25 \mathrm{mg}, 0.110 \mathrm{mmol})$. The reaction was stirred at room temperature for 2 h . The crude product was purified by Biotage chromatography using a gradient of $0-30 \% \mathrm{MeOH}$ in DCM to afford the product as a pale yellow amorphous solid ( $27 \mathrm{mg}, 27 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 11.13(\mathrm{~s}, 1 \mathrm{H}), 10.34(\mathrm{~s}, 1 \mathrm{H}), 10.28(\mathrm{~s}, 1 \mathrm{H}), 8.67(\mathrm{~d}, J=2.1 \mathrm{~Hz}$, $1 \mathrm{H}), 8.50(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.27(\mathrm{dd}, J=8.7,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.16(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.10(\mathrm{~d}, J$ $=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.80(\mathrm{dd}, J=8.5,7.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{dd}, J=8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.72(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $1 \mathrm{H}), 7.57-7.52(\mathrm{~m}, 4 \mathrm{H}), 7.45(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.01(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{dd}, J=12.8$, $5.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.38-4.28(\mathrm{~m}, 6 \mathrm{H}), 3.87-3.76(\mathrm{~m}, 4 \mathrm{H}), 3.65(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 2 \mathrm{H}), 3.54(\mathrm{~s}, 4 \mathrm{H})$, $2.89(\mathrm{ddd}, J=16.9,13.8,5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.69-2.32(\mathrm{~m}, 12 \mathrm{H}), 2.08-1.99(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 172.78,169.93,166.79,165.27,164.98,164.65,155.83,148.27,146.58$, $142.97,138.59,137.56,136.97,134.88,133.23,131.37,129.37,128.83,128.42,127.90,127.30$, $126.23,123.55,121.82,121.33,119.99,119.82,119.24,116.93,116.75,116.29,115.39,79.18$, $70.10,69.70,68.90,68.66,64.42,64.03,57.22,53.03,48.75,30.96,22.02$ (1 signal not observed/overlapping signals). HRMS (ESI $)$ : calcd for $\mathrm{C}_{49} \mathrm{H}_{49}{ }^{35} \mathrm{ClN}_{7} \mathrm{O}_{11}(\mathrm{M}+\mathrm{H})^{+}$947.3173, found 947.3173 .
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-methylquinoline-6-carboxamide 18

2,3-Dihydrobenzo[b][1,4]dioxine-6-carboxylic acid ( $1.27 \mathrm{~g}, 7.06 \mathrm{mmol}$ ) was suspended in anhydrous DCM ( 20 mL ) under inert atmosphere. DMF ( $1.23 \mu \mathrm{l}, 0.016 \mathrm{mmol}$ ) and oxalyl chloride ( $0.65 \mathrm{~mL}, 7.70 \mathrm{mmol}$ ) were added dropwise and the resulting green solution was allowed to stir at $20^{\circ} \mathrm{C}$ for 3 h , after which time it was concentrated under vacuum to afford a dry pale green solid. The solid was dissolved in pyridine ( 20 mL ) and N -(5-amino-2-chlorophenyl)-2-methylquinoline-6-carboxamide 24 ( $2.00 \mathrm{~g}, 6.42 \mathrm{mmol}$ ) was added in one portion. The resulting dark yellow suspension was allowed to stir for 2 h , then it was poured onto water and the yellow precipitate was filtered and washed several times with water, $\mathrm{Et}_{2} \mathrm{O}$ and finally with a minimum amount of DCM to afford the crude product as a pale yellow amorphous solid, which was carried onto the next step without purification ( $1.86 \mathrm{~g}, 61 \%$ ). ${ }^{1} \mathrm{H}-\mathrm{NMR}$ (500 MHz, DMSO- $d_{6}$ ): $\delta 10.31(\mathrm{~s}, 1 \mathrm{H}), 10.27(\mathrm{~s}, 1 \mathrm{H}), 8.63(\mathrm{~d}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.43(\mathrm{~d}, J=8.8 \mathrm{~Hz}$, $1 \mathrm{H}), 8.25(\mathrm{dd}, J=8.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.14(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{dd}, J$ $=8.8,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.58-7.49(\mathrm{~m}, 4 \mathrm{H}), 7.00(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.37-4.26(\mathrm{~m}, 4 \mathrm{H}), 2.71(\mathrm{~s}, 3 \mathrm{H})$. HRMS (ESI ${ }^{+}$): Found $[\mathrm{M}+\mathrm{H}]^{+} 474.1210 \mathrm{C}_{26} \mathrm{H}_{21}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{4}$ requires 474.1215.
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl) -2-(piperazin-1-ylmethyl)quinoline-6-carboxamide 19
$N$-[2-Chloro-5-(2,3-dihydro-1,4-benzodioxine-6-carbonylamino)phenyl)-2-methyl-quinoline-6-carboxamide $18(4.00 \mathrm{~g}, 8.44 \mathrm{mmol})$ was taken up in anhydrous DMF ( 20.00 mL ) and anhydrous 1,4-dioxane ( 20 mL ). Selenium dioxide ( $1.03 \mathrm{~g}, 9.28 \mathrm{mmol}$ ) was added and the
reaction was degassed via 3 x vacuum/nitrogen cycles. The reaction mixture was heated to $50^{\circ} \mathrm{C}$ and stirred for 16 h . Then, the reaction was filtered through Celite, eluting with $1: 1 \mathrm{DCM} / \mathrm{EtOAc}$ and concentrated in vacuo. Remaining DMF was removed by azeotrope with EtOAc/heptane to afford $\quad N$-(2-chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-formylquinoline-6-carboxamide as a brown solid. The crude material was used directly in the next step, assuming 100\% yield, without further purification. Sodium cyanoborohydride ( 2.12 g , $33.8 \mathrm{mmol})$ was added to a stirring suspension of $N$-(2-chloro-5-(2,3-dihydro-1,4-benzodioxine-6-carbonylamino)phenyl)-2-formyl-quinoline-6-carboxamide ( $4.12 \mathrm{~g}, 8.44 \mathrm{mmol}$ ) and 1-Bocpiperazine ( $3.15 \mathrm{~g}, 16.89 \mathrm{mmol}$ ) in anhydrous DMF $(60 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ under nitrogen. Then, acetic acid ( $0.53 \mathrm{~mL}, 9.29 \mathrm{mmol}$ ) was added. The reaction mixture was allowed to warm to room temperature and stirred overnight. A bleach bubbler was used to vent the reaction. The reaction was quenched by slow addition of an aqueous solution of $1 \mathrm{M} \mathrm{NaOH}(50 \mathrm{ml})$ and the reaction mixture concentrated in vacuo, using heptane for azeotropic removal of DMF. The resulting residue was taken up in a small amount of MeOH and poured into a large volume of water (300400 ml ). The precipitate formed was collected by filtration, washed with water, then $\mathrm{Et}_{2} \mathrm{O}$ and dried under vacuum overnight to afford the crude tert-butyl 4-((6-((2-chloro-5-(2,3dihydrobenzo $[b][1,4]$ dioxine-6-carboxamido)phenyl)carbamoyl)quinolin-2-yl)methyl)piperazine-1-carboxylate as a pale brown solid ( 4.57 g ). This material was used directly in the next reaction without further purification.

To a solution of tert-butyl 4-((6-((2-chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)carbamoyl)quinolin-2-yl)methyl)piperazine-1-carboxylate (4.57 g, 6.94 mmol) in anhydrous $\mathrm{MeOH}(60 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ under argon was added 4 M HCl in dioxane (26.1 $\mathrm{mL}, 104.4 \mathrm{mmol}$ ). Upon addition of the acid the reaction mixture became darker red/brown in
color. The reaction was allowed to warm to room temperature and stirred overnight. Then, the solvents were removed in vacuo. The resulting residue was suspended in a small amount of MeOH . The suspension was poured into a large volume of water (with stirring) and the precipitate formed was collected by filtration, washed well with water, then $\mathrm{Et}_{2} \mathrm{O}$ and dried under vacuum to afford the crude product as a brown solid ( 3.42 g ). Purification by Biotage column chromatography using a gradient of $0-10 \% \mathrm{MeOH}$ in $\mathrm{DCM}+1 \% 7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH afforded the title compound as a dark yellow amorphous solid ( $1.53 \mathrm{~g}, 32 \%$ over 3 steps). ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{DMSO}_{6}\right) \delta 10.32(\mathrm{~s}, 1 \mathrm{H}), 10.27(\mathrm{~s}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.49(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $1 \mathrm{H}), 8.26(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.15(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.09(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.76-7.71$ $(\mathrm{m}, 2 \mathrm{H}), 7.58-7.47(\mathrm{~m}, 3 \mathrm{H}), 7.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.40-4.23(\mathrm{~m}, 4 \mathrm{H}), 3.76(\mathrm{~s}, 2 \mathrm{H}), 2.74(\mathrm{t}$, $J=4.8 \mathrm{~Hz}, 4 \mathrm{H}), 2.47-2.42(\mathrm{br} \mathrm{m}, 4 \mathrm{H})\left(1\right.$ proton missing). ${ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $d_{6}$ ) $\delta$ $165.00,164.64,161.79,148.29,146.57,142.97,138.58,137.45,134.89,131.32,129.36,128.81$, $128.39,127.85,127.31,126.21,123.54,121.85,121.32,119.81,119.23,116.92,116.74,65.11$, 64.42, 64.03, 54.26, 45.54. HRMS (ESI $)$ : calcd for $\mathrm{C}_{30} \mathrm{H}_{29}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+} 558.1903$, found 558.1885.

## 2-(2-(2-(4-( (6-)( $2-$ Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-

 carboxamido)phenyl)carbamoyl)quinolin-2-yl)methyl)piperazin-1-yl)ethoxy)ethoxy)ethyl 4methylbenzenesulfonate 20$$
N \text {-(2-Chloro-5-(2,3-dihydrobenzo }[b][1,4] \text { dioxine-6-carboxamido)phenyl) } \quad-2 \text {-(piperazin-1- }
$$ ylmethyl)quinoline-6-carboxamide 19 ( $500 \mathrm{mg}, 0.90 \mathrm{mmol}$ ) was dissolved in anhydrous DMF (8 $\mathrm{ml})$ under nitrogen at room temperature. $\mathrm{K}_{2} \mathrm{CO}_{3}(372 \mathrm{mg}, 2.69 \mathrm{mmol})$ was added, followed by a solution of 2-(2-(2-hydroxyethoxy)ethoxy)ethyl 4-methylbenzenesulfonate ( $573 \mathrm{mg}, 1.88 \mathrm{mmol}$ )

in anhydrous DMF ( 2 mL ) and the reaction stirred at room temperature overnight. Further 2-(2-(2-hydroxyethoxy)ethoxy)ethyl 4-methylbenzenesulfonate ( $286 \mathrm{mg}, 0.94 \mathrm{mmol}$ ) in anhydrous DMF ( 1 mL ) was added and the reaction stirred overnight. The reaction mixture was poured into water and the aqueous layer extracted three times with $10 \% \mathrm{MeOH}$ in DCM . The combined organic layer was washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. EtOAc/heptane was added to remove remaining traces of DMF by azeotrope. The crude oil was purified by Biotage chromatography using a gradient of $0-10 \% \mathrm{MeOH}$ in $\mathrm{DCM}+1 \% 7 \mathrm{NH}_{3}$ in MeOH to afford the title compound as a yellow amorphous solid ( $297 \mathrm{mg}, 48 \%$ ). ${ }^{1} \mathrm{H} \mathrm{NMR}$ ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 8.63(\mathrm{~s}, 1 \mathrm{H}), 8.59(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.42(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.28(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $1 \mathrm{H}), 8.21(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.17(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.98(\mathrm{dd}, J=8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.95$ (s, 1H), $7.76(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\mathrm{~s}, 1 \mathrm{H}), 7.45(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.40(\mathrm{dd}, J=8.4,2.2 \mathrm{~Hz}$, $1 \mathrm{H}), 6.97(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.42-4.43(\mathrm{~m}, 4 \mathrm{H}), 3.90(\mathrm{~s}, 2 \mathrm{H}), 3.78-3.71(\mathrm{~m}, 2 \mathrm{H}), 3.71-3.66$ $(\mathrm{m}, 2 \mathrm{H}), 3.66-3.59(\mathrm{~m}, 6 \mathrm{H}), 2.74-2.56(\mathrm{~m}, 10 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 165.20$, $165.02,162.36,149.26,147.08,143.70,138.10,137.49,134.77,131.89,130.26,129.63,127.91$, 127.88, 126.89, 122.42, 120.64, 117.96, 117.57, 117.05, 116.93, 112.63, 72.76, 70.49, 70.44, 68.81, 65.16, 64.70, 64.32, 61.83, 57.90, 53.69, 53.39. (1 signal not observed/overlapping signals). HRMS (ESI $)$ : calcd for $\mathrm{C}_{36} \mathrm{H}_{41}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+} 690.2689$, found 690.2692 .
$N$-(4-Chloro-3-nitrophenyl)-2,3-dihydrobenzo[ $b][1,4]$ dioxine-5-carboxamide 2,3-Dihydro-1,4-benzodioxine-5-carboxylic acid ( $1.73 \mathrm{~g}, 9.61 \mathrm{mmol}$ ) was suspended in anhydrous DCM ( 45 mL ) and 3 drops of anhydrous DMF were added. Oxalyl chloride ( 0.81 mL , 9.61 mmol ) was added dropwise (effervescence observed) and the reaction stirred at room temperature under nitrogen for 2 h . After this time the effervescence had ceased and the reaction
mixture had become fully dissolved. Then, the solvents were removed in vacuo. Anhydrous DCM ( 10 mL ) was added and the reaction concentrated again. The residue was taken up in anhydrous DCM ( 5 mL , then 1 mL to rinse flask) and added slowly to a solution of 4-chloro-3-nitro-aniline $23(1.11 \mathrm{~g}, 6.41 \mathrm{mmol})$ and pyridine ( 1.55 mL , 19.22 mmol ) in anhydrous DCM (45 $\mathrm{mL})$. The reaction was stirred at room temperature under nitrogen for 48 h . The solvent was removed in vacuo and the resulting residue partitioned between saturated aqueous $\mathrm{NaHCO}_{3}$ solution and $10 \% \mathrm{MeOH}$ in $\mathrm{DCM}(30 \mathrm{~mL})$. The aqueous layer was extracted with two further portions of $10 \% \mathrm{MeOH}$ in $\mathrm{DCM}(30 \mathrm{ml})$, the combined organic layer was washed with water, brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo, then dried in vacuo to afford the crude product as a pale brown solid $(2.24 \mathrm{~g})$.

However, this material was contaminated with 2,3-dihydro-1,4-benzodioxine-5-carboxylic acid, therefore the solid was suspended in a small volume of MeOH and diluted with saturated aqueous $\mathrm{NaHCO}_{3}$ solution. The suspension was vigorously stirred for 1 h , after which time the precipitate was collected by filtration. The precipitate was washed with copious amounts of water, followed by a small portion of $\mathrm{Et}_{2} \mathrm{O}$, then dried under vacuum to afford the product as a beige amorphous solid $(1.82 \mathrm{~g}, 84 \%) .{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}\right.$, DMSO- $\left.d_{6}\right) \delta 10.62(\mathrm{~s}, 1 \mathrm{H}), 8.55(\mathrm{~d}, J$ $=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.95(\mathrm{dd}, J=8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{dd}, J=7.6,1.6 \mathrm{~Hz}$, $1 \mathrm{H}), 7.05(\mathrm{dd}, J=8.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.94(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.39-4.26(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR $(126$ MHz, DMSO- $d_{6}$ ) $\delta 164.71,147.29,143.70,141.31,138.82,132.01,124.79,124.51,121.25$, $120.84,119.59,118.73,115.87,64.52,63.77$. $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right)$: calcd for $\mathrm{C}_{15} \mathrm{H}_{12}{ }^{35} \mathrm{ClN}_{2} \mathrm{O}_{5}(\mathrm{M}+\mathrm{H})^{+}$ 335.0429, found 335.0413.
$N$-(3-Amino-4-chlorophenyl)-2,3-dihydrobenzo[b][1,4]dioxine-5-carboxamide
$N$-(4-chloro-3-nitro-phenyl)-2,3-dihydro-1,4-benzodioxine-5-carboxamide ( $1.50 \mathrm{~g}, 4.48 \mathrm{mmol}$ ) was suspended in a mixture of $\mathrm{DCM}(18 \mathrm{~mL})$ and $\mathrm{EtOH}(9 \mathrm{~mL}) \cdot 10 \% \mathrm{Pd} / \mathrm{C}(200 \mathrm{mg})$ was added and the reaction mixture stirred for 3 days under $1 \mathrm{~atm} \mathrm{H}_{2}$. The reaction mixture was filtered through Celite, eluting with $10 \% \mathrm{MeOH}$ in DCM . The filtrate was concentrated in vacuo and dried under high vacuum to afford the crude product as a pale brown solid ( $1.13 \mathrm{~g}, 93 \%$ purity, $77 \%$ ) which was used directly in the next reaction. A portion of this material ( 320 mg ) was purified by Biotage chromatography using a gradient of 0 to $5 \%$ EtOAc in DCM to afford the title compound as a pale yellow amorphous solid ( $198 \mathrm{mg}, 52 \%$ based on starting material). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 9.93(\mathrm{~s}, 1 \mathrm{H}), 7.35(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 2 \mathrm{H})$, $6.99(\mathrm{dd}, J=8.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.81(\mathrm{dd}, J=8.6,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.37(\mathrm{~s}$, $2 \mathrm{H}), 4.39-4.32(\mathrm{~m}, 2 \mathrm{H}), 4.31-4.27(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO-d ${ }_{6}$ ) $\delta$ 163.77, $144.72,143.63,141.13,138.48,128.82,125.80,121.21,120.70,118.98,111.77,108.78,106.29$, 64.43, 63.74. HRMS (ESI $)$ : calcd for $\mathrm{C}_{15} \mathrm{H}_{14}{ }^{35} \mathrm{ClN}_{2} \mathrm{O}_{3}(\mathrm{M}+\mathrm{H})^{+}$305.0687, found 305.0688.
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-5-carboxamido)phenyl)-2-methylquinoline-6-carboxamide

2-Methylquinoline-6-carboxylic acid ( $685 \mathrm{mg}, 3.66 \mathrm{mmol}$ ) was dissolved in anhydrous DCM $(24 \mathrm{~mL})$ at room temperature under inert atmosphere and 1 drop anhydrous DMF added. Then, oxalyl chloride ( $0.31 \mathrm{~mL}, 3.66 \mathrm{mmol}$ ) was added dropwise (effervescence observed) and the reaction stirred at room temperature for 2 h , over which time effervescence ceased and the solution became dark green. The solvent was removed in vacuo. The resulting residue was redissolved in anhydrous DCM ( 2 ml ) and concentrated in vacuo (x 2). The solid was taken up in anhydrous DCM ( 1 ml , then 1 ml to rinse flask) and slowly added to a stirring solution of N -(3-
amino-4-chloro-phenyl)-2,3-dihydro-1,4-benzodioxine-5-carboxamide ( $800.00 \mathrm{mg}, 2.44 \mathrm{mmol}$ ) in anhydrous pyridine ( 18 mL ) at room temperature under inert atmosphere. The reaction was stirred overnight. A precipitate formed which was isolated by filtration, washed with water (x 2), ether and then dried under vacuum to afford the product as a beige amorphous solid ( 672 mg , $58 \%) .{ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 10.33(\mathrm{~s}, 1 \mathrm{H}), 10.32(\mathrm{~s}, 1 \mathrm{H}), 8.63(\mathrm{~d}, J=1.8 \mathrm{~Hz}, 1 \mathrm{H})$, $8.42(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.25(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.07(\mathrm{~d}, J=2.3 \mathrm{~Hz}, 1 \mathrm{H}), 8.04(\mathrm{~d}, J=8.8$ $\mathrm{Hz}, 1 \mathrm{H}), 7.68(\mathrm{dd}, J=8.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{dd}, J=8.6,2.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.15(\mathrm{dd}, J=7.6,1.5 \mathrm{~Hz}$, $1 \mathrm{H}), 7.02(\mathrm{dd}, J=8.0,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.37(\mathrm{dd}, J=5.1,2.4 \mathrm{~Hz}, 2 \mathrm{H}), 4.31$ $(\mathrm{dd}, J=5.0,2.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.71(\mathrm{~s}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 126 MHz, DMSO- $d_{6}$ ) $\delta$ 165.07, 164.24, 160.97, 148.57, 143.67, 141.20, 138.27, 137.25, 135.12, 130.87, 129.53, 128.46, 128.44, 127.85, 125.51, 125.39, 123.79, 123.07, 121.19, 120.76, 119.31, 119.19, 118.67, 64.47, 63.77, 25.05. HRMS (ESI $)$ : calcd for $\mathrm{C}_{26} \mathrm{H}_{21}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+} 474.1215$, found 474.1192.

## $N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-5-carboxamido)phenyl)-2-(piperazin-1-

 ylmethyl)quinoline-6-carboxamide$N$-(2-Chloro-5-(2,3-dihydro-1,4-benzodioxine-5-carbonylamino)phenyl)-2-methyl-quinoline-6-carboxamide ( $500 \mathrm{mg}, 1.06 \mathrm{mmol}$ ) was taken up in anhydrous DMF ( 5 mL ) and anhydrous 1,4-dioxane ( 5 mL ). Selenium dioxide ( $128 \mathrm{mg}, 1.16 \mathrm{mmol}$ ) was added, the reaction was degassed via 3 x vacuum/nitrogen cycles and stirred at $50^{\circ} \mathrm{C}$ for 5 h . The reaction mixture was filtered through Celite (eluting with $1: 1 \mathrm{DCM} / \mathrm{EtOAc}$ ), then concentrated in vacuo, using EtOAc/heptane to remove traces of DMF. This material was used directly in the next reaction without further purification, assuming $100 \%$ yield.
$N$-[2-Chloro-5-(2,3-dihydro-1,4-benzodioxine-5-carbonylamino)phenyl]-2-formyl-quinoline-6carboxamide ( $0.51 \mathrm{~g}, 1.06 \mathrm{mmol}$ ) and 1-Boc-piperazine ( $392 \mathrm{mg}, 2.11 \mathrm{mmol}$ ) were dissolved in anhydrous DMF $(10 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ under nitrogen. Then, sodium cyanoborohydride ( $265 \mathrm{mg}, 4.22$ mmol ) was added in one portion, followed by acetic acid ( $0.07 \mathrm{~mL}, 1.16 \mathrm{mmol}$ ). A bleach bubbler was used to vent the reaction. The reaction was allowed to warm to room temperature and stirred overnight. The reaction was quenched by addition of 1 M NaOH (aq.) and then concentrated in vacuo. The resulting residue was suspended in MeOH and diluted with water to afford a brown precipitate. The precipitate was collected by filtration and washed well with water, followed by $\mathrm{Et}_{2} \mathrm{O}$ and dried under vacuum to afford the product as a pale brown solid (520 mg ). This material was used directly in the next reaction without further purification.
tert-Butyl 4-((6-((2-chloro-5-(2,3-dihydro-1,4-benzodioxine-5
carbonylamino)phenyl)carbamoyl)-2-quinolyl)methyl)piperazine-1-carboxylate ( $508 \mathrm{mg}, 0.770$ mmol ) was dissolved in anhydrous $\mathrm{MeOH}(7 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ under inert atmosphere. Then, 4 M HCl in dioxane ( $2.90 \mathrm{~mL}, 11.61 \mathrm{mmol}$ ) was added and the reaction warmed to room temperature. After 1.5 h , further 4 M HCl in dioxane ( $2.90 \mathrm{~mL}, 11.61 \mathrm{mmol}$ ) was added and the reaction stirred overnight. Then, the reaction mixture was concentrated in vacuo. The resulting residue was taken up in $1: 1 \mathrm{MeOH} / \mathrm{DCM}$ and subjected to Isolute Flash SCX-II chromatography to afford the free salt, eluting first with MeOH followed by $10 \% 7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH . This material was further purified by Biotage chromatography using a gradient of $0-10 \% \mathrm{MeOH}$ in $\mathrm{DCM}+1 \%$ $7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH to afford the title compound as a pale brown amorphous solid ( $253 \mathrm{mg}, 43 \%$ over 3 steps). ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 10.35(\mathrm{~s}, 1 \mathrm{H}), 10.32(\mathrm{~s}, 1 \mathrm{H}), 8.71-8.54(\mathrm{~m}$, $1 \mathrm{H}), 8.49(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.26(\mathrm{dd}, J=8.8,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.09$ (app. d, $J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.74$ (d, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.68(\mathrm{dd}, J=8.8,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{dd}, J=7.6,1.3$
$\mathrm{Hz}, 1 \mathrm{H}), 7.02(\mathrm{dd}, J=8.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{t}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.42-4.34(\mathrm{~m}, 2 \mathrm{H}), 4.33-4.29$ $(\mathrm{m}, 2 \mathrm{H}), 3.77(\mathrm{~s}, 2 \mathrm{H}), 2.76(\mathrm{t}, J=4.5 \mathrm{~Hz}, 4 \mathrm{H}), 2.42(\mathrm{br} \mathrm{s}, 4 \mathrm{H})\left(1\right.$ proton missing). ${ }^{13} \mathrm{C}$ NMR (126 $\left.\mathrm{MHz}, \mathrm{DMSO}-d_{6}\right) \delta 165.00,164.24,161.72,148.30,143.67,141.20,138.27,137.48,135.08$, $131.28,129.54,128.81,128.44,127.88,126.22,125.50,123.78,121.86,121.19,120.76,119.32$, $119.19,118.70,65.02,64.47,63.77,53.94,48.60,45.37$ ( 1 missing/overlapping signal). HRMS (ESI ${ }^{+}$): calcd for $\mathrm{C}_{30} \mathrm{H}_{29}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+} 558.1903$, found 558.1890.

## $N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-5-carboxamido)phenyl)-2-((4-(2-(2-(2-

 hydroxyethoxy)ethoxy)ethyl)piperazin-1-yl)methyl)quinoline-6-carboxamide$$
N \text {-(2-Chloro-5-(2,3-dihydro-1,4-benzodioxine-5-carbonylamino)phenyl)-2-(piperazin-1- }
$$ ylmethyl)quinoline-6-carboxamide ( $248 \mathrm{mg}, 0.440 \mathrm{mmol}$ ) was dissolved in anhydrous DMF (3 $\mathrm{ml})$ at room temperature under inert atmosphere and $\mathrm{K}_{2} \mathrm{CO}_{3}(371 \mathrm{mg}, 2.69 \mathrm{mmol})$ was added. Then, a solution of 2-(2-(2 hydroxyethoxy)ethoxy)ethyl 4-methylbenzenesulfonate ( $573 \mathrm{mg}, 1.88$ mmol ) in anhydrous DMF ( 1 mL ) was added and reaction stirred overnight. The reaction was diluted with water and the aqueous layer extracted ( $3 \times 10 \% \mathrm{MeOH}$ in DCM). The combined organic layer was washed with brine and dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$. Purification by Biotage chromatography using a gradient of $0-10 \% \mathrm{MeOH}$ in $\mathrm{DCM}+1 \% 7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH gave the title compound as a pale yellow amorphous solid (108 mg, 35\%). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $9.59(\mathrm{~s}, 1 \mathrm{H}), 8.63(\mathrm{~s}, 1 \mathrm{H}), 8.54(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.42(\mathrm{~d}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.27(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $1 \mathrm{H}), 8.21(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.17(\mathrm{dd}, J=8.8,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.00(\mathrm{dd}, J=8.8,2.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.81$ (dd, $J=7.8,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.75(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.43(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.07(\mathrm{dd}, J=8.0,1.7$ $\mathrm{Hz}, 1 \mathrm{H}), 6.99(\mathrm{t}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.56(\mathrm{dd}, J=5.0,3.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.38(\mathrm{dd}, J=5.0,3.1 \mathrm{~Hz}, 2 \mathrm{H})$, $3.89(\mathrm{~s}, 2 \mathrm{H}), 3.77-3.71(\mathrm{~m}, 2 \mathrm{H}), 3.70-3.58(\mathrm{~m}, 8 \mathrm{H}), 2.73-2.53(\mathrm{~m}, 10 \mathrm{H})(1$ proton missing $)$.

${ }^{13} \mathrm{C}$ NMR $\left(126 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 164.98,163.11,162.34,149.27,143.81,141.99,138.12,137.44$, $134.75,131.96,130.31,129.55,127.80,126.89,124.52,122.41,122.16,121.76,121.56,118.01$, $117.66,113.13,72.74,70.48,70.44,68.82,65.46,65.14,63.75,61.82,57.87,53.67,53.36$ ( 1 overlapping/missing signal). $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right)$: calcd for $\mathrm{C}_{36} \mathrm{H}_{41}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{7}(\mathrm{M}+\mathrm{H})^{+} 690.2689$, found 690.2696.
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-5-carboxamido)phenyl)-2-((4-(2-(2-(2-((2-(2,6-dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-yl)oxy)ethoxy)ethoxy)ethyl)piperazin-1-yl)methyl)quinoline-6-carboxamide 21

2-(2,6-Dioxo-3-piperidyl)-4-hydroxy-isoindoline-1,3-dione (19.87 mg, 0.0700 mmol ) was dissolved in anhydrous THF ( 0.72 mL ) under inert atmosphere, then triphenylphosphine (19.95 $\mathrm{mg}, 0.0800 \mathrm{mmol}$ ) and tert-butyl (NE)-N-tert-butoxycarbonyliminocarbamate ( $17.51 \mathrm{mg}, 0.0800$ mmol) were added, followed by $N$-(2-chloro-5-(2,3-dihydro-1,4-benzodioxine-5-carbonylamino)phenyl)-2-((4-(2-(2-(2-hydroxyethoxy)ethoxy]ethyl)piperazin-1-yl)methyl)quinoline-6-carboxamide ( $50.00 \mathrm{mg}, 0.070 \mathrm{mmol}$ ). The reaction was stirred at room temperature for 2 h . The reaction mixture was pre-absorbed onto silica and purified by Biotage chromatography using a gradient of $0-30 \% \mathrm{MeOH}$ in DCM to afford the product as a pale yellow amorphous solid ( $28 \mathrm{mg}, 41 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 500 MHz, DMSO- $d_{6}$ ) $\delta 11.12(\mathrm{~s}, 1 \mathrm{H}), 10.35(\mathrm{~s}, 1 \mathrm{H})$, $10.32(\mathrm{~s}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.49(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 8.26(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $8.13-8.06(\mathrm{~m}, 2 \mathrm{H}), 7.79(\mathrm{dd}, J=8.6,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.75-7.65(\mathrm{~m}, 2 \mathrm{H}), 7.53(\mathrm{t}, J=8.4 \mathrm{~Hz}, 2 \mathrm{H})$, $7.44(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.14(\mathrm{dd}, J=7.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.02(\mathrm{dd}, J=8.0,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.93(\mathrm{t}, J$ $=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.09(\mathrm{dd}, J=12.8,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.42-4.26(\mathrm{~m}, 6 \mathrm{H}), 3.85-3.73(\mathrm{~m}, 4 \mathrm{H}), 3.64$ (dd, $J=5.9,3.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.54-3.46(\mathrm{dt}, J=11.8,5.7 \mathrm{~Hz}, 4 \mathrm{H}), 2.88(\mathrm{ddd}, J=16.9,13.8,5.4 \mathrm{~Hz}$,
$1 \mathrm{H}), 2.65-2.23(\mathrm{~m}, 12 \mathrm{H}), 2.09-1.96(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (126 MHz, DMSO- $\left.d_{6}\right) \delta$ 172.76, $169.90,166.79,165.24,164.98,164.23,161.71,155.84,148.28,143.66,141.19,138.26,137.51$, $136.95,135.07,133.23,131.27,129.53,128.81,128.43,127.87,126.21,125.50,123.76,121.79$, $121.18,120.76,120.01,119.30,119.18,118.68,116.31,115.37,70.12,69.72,68.90,68.67$, $68.21,64.46,64.35,63.76,57.22,54.91,53.13,53.01,48.74,40.02,30.95,22.01$. HRMS (ESI $^{+}$): calcd for $\mathrm{C}_{49} \mathrm{H}_{49}{ }^{35} \mathrm{ClN}_{7} \mathrm{O}_{11}(\mathrm{M}+\mathrm{H})^{+} 946.3173$, found 946.3183.
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-((4-ethylpiperazin-1-yl)methyl)quinoline-6-carboxamide 22
$N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-methylquinoline-6-carboxamide $\mathbf{1 8}(1.00 \mathrm{~g}, 2.11 \mathrm{mmol})$ was taken up in anhydrous DMF ( 10 mL ) and anhydrous 1,4-dioxane ( 10 mL ). Selenium dioxide ( $280 \mathrm{mg}, 2.53 \mathrm{mmol}$ ) was added, the reaction was degassed via 3 x vacuum/nitrogen cycles and stirred at $50^{\circ} \mathrm{C}$ under nitrogen for 3.5 h . Further selenium dioxide ( $47 \mathrm{mg}, 0.42 \mathrm{mmol}$ ) was added and the reaction stirred at $50{ }^{\circ} \mathrm{C}$ for an additional 2 h . The reaction mixture was cooled, filtered through Celite (eluting with $\mathrm{DCM} / \mathrm{EtOAc}$ ) and concentrated in vacuo, using EtOAc/heptane azeotrope to remove DMF. This material was used directly in the next reaction without further purification, assuming $100 \%$ yield. $N$-(2-Chloro-5-(2,3-dihydrobenzo[b][1,4]dioxine-6-carboxamido)phenyl)-2-formylquinoline-6-carboxamide ( $1.03 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) was suspended in anhydrous DCM ( 20 mL ) at room temperature under nitrogen and 1-ethylpiperazine $(0.80 \mathrm{~mL}, 6.33 \mathrm{mmol})$ was added. The reaction was allowed to stir overnight, then $\mathrm{NaBH}(\mathrm{OAc})_{3}(1.34 \mathrm{~g}, 6.33 \mathrm{mmol})$ was added and stirred for 3 h. The reaction mixture was quenched with a saturated aqueous solution of $\mathrm{NaHCO}_{3}$ and the aqueous layer extracted with $10 \% \mathrm{MeOH}$ in DCM (x 3). The combined organic layer was dried
$\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by Biotage chromatography using a gradient of $0-10 \% \mathrm{MeOH}$ in DCM . This material was further purified by Isolute Flash SCX-II chromatography, eluting with MeOH , followed by $10 \% 7 \mathrm{~N} \mathrm{NH}_{3}$ in MeOH to afford the product as an amorphous yellow solid ( $431 \mathrm{mg}, 35 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 10.32(\mathrm{~s}, 1 \mathrm{H})$, $10.27(\mathrm{~s}, 1 \mathrm{H}), 8.65(\mathrm{~d}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 8.50(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.26(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $8.15(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.09(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.77-7.74(\mathrm{~m}, 1 \mathrm{H}), 7.73(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.56-7.50(\mathrm{~m}, 3 \mathrm{H}), 7.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.34-4.26(\mathrm{~m}, 4 \mathrm{H}), 3.80(\mathrm{~s}, 2 \mathrm{H}), 2.60-2.25(\mathrm{~m}$, $10 \mathrm{H}), 0.99(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $126 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 165.00$, 164.64, 161.70, $148.28,146.57,142.97,138.58,137.52,134.89,131.35,129.35,128.82,128.39,127.87,127.31$, $126.23,123.53,121.82,121.32,119.80,119.23,116.92,116.74,64.42,64.35,64.03,52.94$, 52.34, 51.59, 11.92. HRMS (ESI ${ }^{+}$: calcd for $\mathrm{C}_{32} \mathrm{H}_{33}{ }^{35} \mathrm{ClN}_{5} \mathrm{O}_{4}(\mathrm{M}+\mathrm{H})^{+}$, 586.2216; found 586.2239 .

## $N$-(5-Amino-2-chlorophenyl)-2-methylquinoline-6-carboxamide $\mathbf{2 4}$

$N$-(2-Chloro-5-nitrophenyl)-2-methylquinoline-6-carboxamide was suspended in water (7 mL) and EtOH ( 21 mL ). Ammonium chloride $(2.41 \mathrm{~g}, 45.1 \mathrm{mmol})$ and iron powder ( $2.52 \mathrm{~g}, 45.1$ mmol ) were added and the resulting suspension was allowed to stir at $90^{\circ} \mathrm{C}$ for 1 h . The reaction mixture was allowed to cool to room temperature, diluted with MeOH and DCM and filtered through a pad of Celite. The resulting filtrate was concentrated under vacuum to afford a light brown amorphous solid as crude product, which was used directly in the next step without any further purification $(2.00 \mathrm{~g}, 100 \%) .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 9.96(\mathrm{~s}, 1 \mathrm{H}), 8.58(\mathrm{~d}, J=$ $2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.41(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 8.21(\mathrm{dd}, J=8.7,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 8.02(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H})$, $7.53(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.15(\mathrm{~d}, J=8.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.87(\mathrm{~d}, J=2.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.50(\mathrm{dd}, J=8.7,2.2$
$\mathrm{Hz}, 1 \mathrm{H}), 5.41(\mathrm{bs}, 2 \mathrm{H}), 2.70(\mathrm{~s}, 3 \mathrm{H}) . \mathrm{HRMS}\left(\mathrm{ESI}^{+}\right)$: Found $[\mathrm{M}+\mathrm{H}]^{+} 312.0902 \mathrm{C}_{17} \mathrm{H}_{15}{ }^{35} \mathrm{ClN}_{3} \mathrm{O}$ requires 312.0898 .

## 1-(6-((2-(2-(2-((2-(2,6-Dioxopiperidin-3-yl)-1,3-dioxoisoindolin-4-

yl)oxy)acetamido)ethoxy)ethyl)amino)-6-oxohexyl)-3,3-dimethyl-2-(( $1 E, 3 E)-5-((E)-1,3,3-$
trimethyl-5-sulfoindolin-2-ylidene)penta-1,3-dien-1-yl)-3 H -indol-1-ium-5-sulfonate
In the absence of light, $N$-(2-(2-aminoethoxy)ethyl)-2-(2-(2,6-dioxo-3-piperidyl)-1,3-dioxo-isoindolin-4-yl)oxy-acetamide hydrochloride $4(0.60 \mathrm{mg}, 0.0013 \mathrm{mmol})$ was added to a solution of sodium (2E)-2-((2E,4E)-5-(1-(6-(2,5-dioxopyrrolidin-1-yl)oxy-6-oxo-hexyl)-3,3-dimethyl-5-sulfonato-indol-1-ium-2-yl)penta-2,4-dienylidene)-1,3,3-trimethyl-indoline-5-sulfonate $\mathrm{mg}, 0.0013 \mathrm{mmol}$ ) dissolved in $20 \mu \mathrm{~L}$ of triethylamine and $300 \mu \mathrm{~L}$ of DMF and the deep blue solution was left to stand for for 16 hours. The crude product was then purified by semipreparative HPLC to give the desired product. LCMS $\left(\mathrm{ESI}^{+}\right) \mathrm{RT}=2.59 \mathrm{~min}, 77 \%, \mathrm{M}+\mathrm{H}^{+} 1043$.

## ASSOCIATED CONTENT

## Supporting Information.

The Supporting Information is available free of charge on the ACS Publications website at DOI:
Full chemistry experimental, NMR spectra of final compounds, physicochemical properties and PDP stability experimental, PDP design pictures, biochemical experimental, biology experimental, proteomics experimental, additional information, proteomics data (PDF). SMILES molecular formula strings (CSV)

AUTHOR INFORMATION

## Corresponding Author

*Email: paul.clarke@icr.ac.uk, paul.workman@icr.ac.uk, matthew.cheeseman@icr.ac.uk, phone $(+44) 2087224168$ or keith.jones@icr.ac.uk.

## ORCID

Nicola E. A. Chessum: 0000-0003-4125-320X
A. Elisa Pasqua: 0000-0002-7966-4672

Birgit Wilding: 0000-0002-1896-3708
Ian Collins: 0000-0002-8143-8498

Bugra Ozer: 0000-0002-1441-4162
Mark Stubbs: 0000-0001-7855-9435
Rosemary Burke: 0000-0002-0011-9701
Paul A Clarke: 0000-0001-9342-1290
Paul Workman: 0000-0003-1659-3034

Matthew D. Cheeseman: 0000-0003-1121-6985
Keith Jones: 0000-0002-9440-4094

## Author Contributions

All authors have given approval to the final version of the manuscript. \#These authors contributed equally. S.Y.S. carried out in vitro cellular biology experiments. N.E.A.C and J.J.C. synthesized compounds. N.E.A.C., J.J.C., A.E.P., B.W., G.C., I.C., M.D.C. and K.J. contributed to the design of compounds. M.R. ran and designed the physicochemical and stability analysis of compounds. P.C.M. expressed and purified the recombinant proteins. B.O. analyzed proteomics data. M.R., M.S. and R.B. designed and carried out biochemical experiments. N.E.A.C., S.Y.S.,
P.C., P.W., M.D.C. and K.J. designed studies and interpreted results. N.E.A.C. and M.D.C. wrote the manuscript.

## Notes

The Institute of Cancer Research has a potential financial interest in ligands of pirin and operates a Rewards to Discoverers scheme.

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## ABBREVIATIONS

SAR, structure activity relationships; CETSA, cellular thermal shift assay; ABPP, activitybased protein profiling; PROTAC, proteolysis targeting chimera; SNIPER, specific and nongenetic IAP-dependent protein eraser; PDP, protein degradation probe; SPR, surface plasmon resonance spectroscopy; SEM, standard error of the mean; HPLC, high-performance liquid chromatography; KS, kinetic solubility; tPSA, topological polar surface area; DMSO, dimethyl sulfoxide; HSF1, heat shock transcription factor 1; BRD4, bromodomain-containing protein 4; VHL, Von Hippel-Lindau disease tumor suppressor; IAP, inhibitors of apoptosis proteins; CRBN, protein cereblon; PPI, protein-protein interaction; TBAF, tetra- $n$-butylammonium fluoride; HBD, hydrogen bond donor; TMT, tandem mass tagging; MS2, MS/MS mass spectrometry; DTBAD, di-tertbutylazocarboxylate; THF, tetrahydrofuran; DCM,
dichloromethane; RT, room temperature; HATU, 1-[bis(dimethylamino)methylene]-1 $\mathrm{H}-1,2,3-$ triazolo[4,5-b]pyridinium 3-oxid hexafluorophosphate; DIPEA, diisopropylethylamine; DMF, dimethylformamide; DBU, 1,8-diazabicyclo(5.4.0)undec-7-ene; SF, significant figures.

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Table of Contents graphic



Figure 1. (Top) Pirin (5JCT)/CRBN (5CI1)/PDP ternary complex design model. The PDP can either stabilize a PPI or simply bring the proteins in close proximity, depending on the role of the linker. (Bottom Left) Cartoon representation of the chemical probe 1 (yellow) bound to recombinant pirin (5JCT). The cloud represents the shape of the binding pocket with key residues shown in black and the metal in orange.
Red=oxygen, blue=nitrogen. Hydrogens and solvent are omitted for clarity except the water coordinated to the metal, shown as a red sphere. Both representations were generated using the PyMOL Molecular Graphics System, Version 1.8 Schrödinger, LLC. (Bottom Right) Key residues in the binding site and the clear solvent exposed vector for the chemical probe 1 binding to pirin are shown, adapted from an analysis using MOE 2014.09. The ethyl pyrrolidine solubilizing group of chemical probe 1 was not resolved in the crystal structure and therefore is not drawn in the analysis.
$1510 \times 901 \mathrm{~mm}(96 \times 96$ DPI)


Table 1. Physicochemical Properties and Affinities for Recombinant Protein Targets of the Three Generations of PDPs
$\mathrm{SF}=$ significant figure. All data was reprocessed using GraphPad Prism 7.01. See figures S8-S16.
SEM=standard error of the mean. aHBD=hydrogen bond donor count. bALogP was calculated using Biovia Pipeline Pilot Version 9.5, 2 SF. cLogD7.4 measured using a HPLC-based method, $n=1,2$ SF. dtPSA was calculated using ChemDraw (16.0.1.4) based on the O- and N-count, 3 SF . eKS=kinetic solubility in pH7.4 phosphate buffer at room temperature, $\mathrm{n}=1,1 \mathrm{SF}$. fKD values are reported to 2 SF and are calculated by equilibrium analysis using a one site specific binding model from SPR sensorgrams at equilibrium where possible, $\mathrm{pKD}=-\log (\mathrm{KD}(\mathrm{M}) \times 10-9)$ and represents the geometric mean of $\mathrm{n}=3$ independent biological repeats. gIC50 values are reported to 2 SF and are calculated from an FP-assay dose-response curve to displace a thalidomide derived fluorescent probe using a log[Inhibitor] vs. response - Variable slope (four parameters) model, pIC50 $=-\log$ (IC50 (M) $\times 10-9$ ) and represents the geometric mean of $n=3$ independent biological repeats, also see reference 31 . hKi values are calculated from the geometric mean CRBN-DDB1 complex IC50 and the FP-probe KD using methods described in reference 32.

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755 \times 203 \mathrm{~mm}(96 \times 96 \text { DPI })
$$



Reagents and conditions: (a) i) PPh3, tButyl 2-hydroxyacetate, DTBAD, THF, $0^{\circ} \mathrm{C} \rightarrow \mathrm{RT}, 16 \mathrm{~h}, 75 \%$; ii) HCO2H, DCM, $40^{\circ} \mathrm{C}, 16 \mathrm{~h}, 54 \%$; iii) HATU, DIPEA, DMF, tButyl (2-(2-aminoethoxy)ethyl)carbamate, RT, 16 h, $81 \%$; iv) 4 M HCl in dioxane, $0.5 \mathrm{~h}, 100 \%$; (b) i) ethyl 4-bromobutanoate, K2CO3, DMF, RT, 16 h, 37\%;
(c) Herrmann's palladacycle,29 tBu3PHBF4, MoCO6, DBU, 2-(trimethylsilyl)ethan-1-ol, 130 oC, 62\% (d) i) TBAF, THF, RT, 16 h; (e) i) HATU, DIPEA, DMF, RT, 38\% (over 2 steps); ii) LiOH.H2O, MeOH/THF/H2O, RT, 48 h, 18\% (f) HATU, DIPEA, DMF, 74\%. For the synthesis of hydroxythalidomide 2 see reference 28.

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Reagents and conditions: (a) i) $X=F$ 12, 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM; RT, 3 h; then pyridine, 18 h Quant; ii) $\mathrm{Fe}(0), \mathrm{NH} 4 \mathrm{Cl}, \mathrm{EtOH} / \mathrm{H} 2 \mathrm{O}, 90^{\circ} \mathrm{C}, 1 \mathrm{~h}$, Quant; (b) 2,3-dihydrobenzo-[b][1,4]-dioxine-6-carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine RT, $2 \mathrm{~h}, 85 \%$; (c) i) SeO2, 1,4-dioxane/DMF, reflux, 1 h ; ii) N-Boc-piperazine, DCM, RT, 12 h then $\mathrm{NaBH}(\mathrm{OAc}) 3, \mathrm{DCM}, \mathrm{RT}, 2 \mathrm{~h}$, 94\% (over 2 steps); iii) TFA, DCM, RT, 2 h, 69\% (d) i) 2, PPh3, tButyl 2-hydroxyacetate, DTBAD, THF, 0
${ }^{\circ} \mathrm{C} \rightarrow$ RT, 16 h, 75\%; ii) HCO2H, DCM, $40^{\circ} \mathrm{C}, 16 \mathrm{~h}, 54 \%$; iii) HATU, DIPEA, DMF, tButyl 3-(2aminoethoxy)propanoate, RT, 16 h, $72 \%$; iv) HCO2H, DCM, $40^{\circ} \mathrm{C}, 6 \mathrm{~h}, 93 \%$ (e) HATU, DIPEA, DMF, RT, 16 $\mathrm{h}, 52 \%$; (f) i) $\mathrm{X}=\mathrm{Cl} 17$, 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine $2 \mathrm{~h}, 88 \%$; ii) $\mathrm{Fe}(0), \mathrm{NH} 4 \mathrm{Cl}, \mathrm{EtOH} / \mathrm{H} 2 \mathrm{O}, 90^{\circ} \mathrm{C}, 1 \mathrm{~h}$, Quant; (g) i) 2,3-dihydrobenzo-[b][1,4]-dioxine-6carboxylic acid, oxalyl chloride, DMF, DCM, RT, 3 h ; then pyridine $2 \mathrm{~h}, 61 \%$; (h) i) SeO2, DMF, 1,4-dioxane, $50^{\circ} \mathrm{C}, 16 \mathrm{~h}$; ii) N -Boc-piperazine, $\mathrm{NaBH} 3 \mathrm{CN}, \mathrm{AcOH}, \mathrm{DMF}, 0^{\circ} \mathrm{C} \rightarrow \mathrm{RT}, 16 \mathrm{~h}$ iii) 4 M HCl in dioxane, MeOH 0 ${ }^{\circ} \mathrm{C} \rightarrow$ RT, 16 h, 32\% over 3 steps; (i) 2-[2-(2-hydroxyethoxy)ethoxy]ethyl 4-methylbenzenesulfonate, K2CO3, DMF, RT, 16 h, 48\%; j) PPh3, DTBAD, THF, RT, 2 h, 27\%; (k) i) 23, 2,3-dihydro-1,4-benzodioxine-5-carboxylic acid, oxalyl chloride, DMF, DCM, RT, 2 h ; then pyridine $48 \mathrm{~h}, 84 \%$; ii) Pd/C, EtOH/DCM, H2 (1 atm), 77\%; iii) 2-methylquinoline carboxylic acid, oxalyl chloride, DMF, DCM; RT, 2 h ; then pyridine 16 h , $58 \%$; then same procedure as from $18,6 \%$ yield over 5 steps. (I) SeO2, 1,4-dioxane/DMF, $50^{\circ} \mathrm{C}, 5.5 \mathrm{~h}$; ii) N-ethylpiperazine, DCM, RT, 20 h ; then $\mathrm{NaBH}(\mathrm{OAc}) 3, \mathrm{DCM}, \mathrm{RT}, 2 \mathrm{~h}, 35 \%$ (over 2 steps).


Figure 2. (A) Representative SPR sensorgram of the third generation PDP 16 and recombinant pirin. (B) Representative binding curve of PDP 16 in the CRBN-DDB1 FP-assay.
$1363 \times 457 \mathrm{~mm}(96 \times 96$ DPI)


Figure 3. A: Immunoblot of SK-OV-3 human ovarian cancer cells demonstrating the depletion of pirin protein using the third generation PDP 16 and the time-dependent hook-effect. B: Immunoblot demonstrating the concentration-dependent depletion of pirin protein after 2 h exposure in SK-OV-3 cells. C: Capillary electrophoresis and immunoassay were used to quantify the pirin protein expression after 2 h exposure with PDP 16, all values are normalized to vinculin loading control and relative to the measured basal pirin protein expression, all bars represent the arithmetic mean of $n=3$ independent biological repeats, error bars are
SEM. D: Proteomics analysis of the third generation PDP 16 ( 50 nM ) exposure ( 4 h ) in SK-OV-3 cells compared to vehicle control, using a tandem mass tagging (TMT) MS2 protocol on the cell lysate, 8547 quantifiable proteins were identified, each blue dot represents a single quantifiable protein, pirin is marked in red (adjusted $p$ value $=1.4 \times 10-4$ ), $p$ values were calculated using a linear modelling based $t$-test and corrected for multiple comparisons using the Benjamini-Hochberg method to give the $\mathrm{p}(\mathrm{adj})$ values shown, dotted lines represent 2 -fold depletion of the protein and a $p(a d j)=0.05$.


Figure 4. Intracellular competition studies with PDP 16 and the chemical probes. SK-OV-3 cells were pretreated with increasing concentrations of chemical probe for 4 h before exposing to PDP 16 for 2 h at the concentrations shown. Cells were then lysed and protein expression analyzed using immunoblot. For clarity, gel images have been cropped where appropriate.
$143 \times 107 \mathrm{~mm}(150 \times 150$ DPI)

